

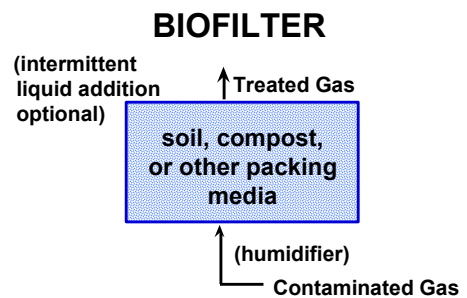
BIODEGRADATION OF BTEX-CONTAMINATED GAS IN A SPARGED SHALLOW LIQUID REACTOR

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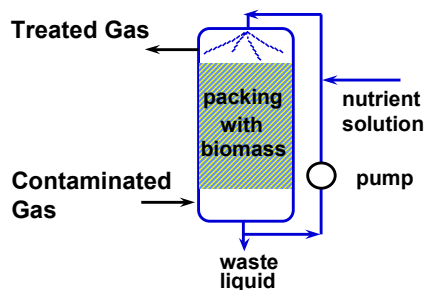
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- Sources of contaminated gases
 - industrial emissions
 - SVE off-gas from site-remediation

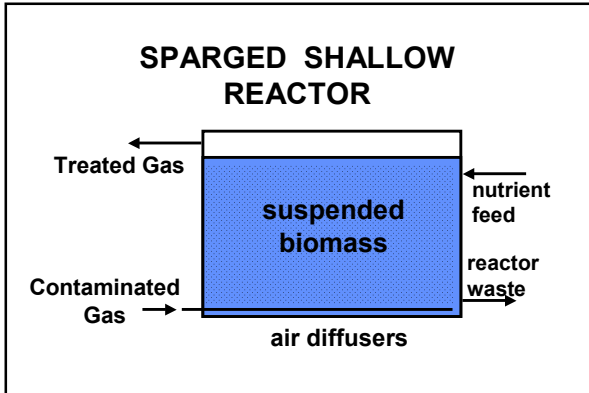
- Conventional gas treatment methods
 - activated carbon
 - incineration
- Biological gas treatment reactors



BIOTRICKLING FILTER



- BTEX treatment demonstrated
- Excess biogrowth affects reliability
 - potential media plugging
 - potential gas short circuiting



DESIGN AND OPERATION OF A SPARGED REACTOR

- What liquid depth is required?
- What operating SRT ?

OBJECTIVES

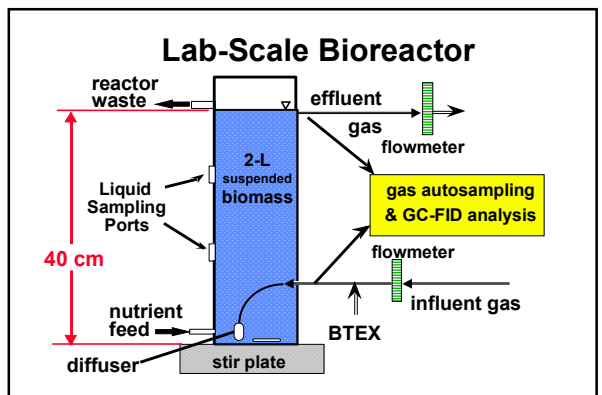
- Investigate the ability to achieve high gas treatment efficiency for BTEX compounds
- Investigate the effect of operating SRT on gas treatment efficiency

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DEVELOP A MASS TRANSFER & BIODEGRADATION MODEL

- BTEX Mass Transfer Experiments
 - Henry's coefficient
 - relate BTEX & oxygen mass transfer coefficients ($K_L a$)

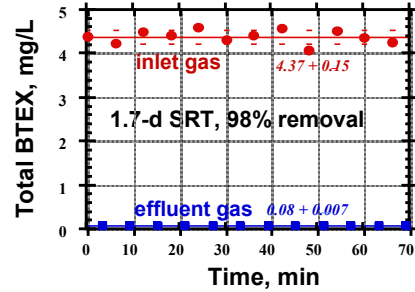
- BTEX Biodegradation Kinetics
 - Michaelis Menten model for individual compounds (K & K_s)
 - effect of mixtures
 - competitive inhibition model



Key Operating Conditions

- Liquid Depth = 40 cm
- Liquid Volume = 2 L
- SRT = 1.7, 2.7, 9.2 days
- Oxygen K_{La} = 0.1 to 0.2 min^{-1}
- Temperature = 20 to 24 °C
- Q_g / A = 4 to 8 cm/min
- Inlet Gas BTEX 2.3 - 4.3 mg/L

BTEX Removal Over Analytic Period



Average LongTerm Gas Treatment Results

SRT, days	Load, mg/L-hr	VSS mg/L	BTEX, mg/L	
			Inlet	Effluent
1.7	15.3	444	4.32	0.01
2.7	16.8	756	2.92	<0.01
9.2	14.9	1327	2.29	0.04

BTEX Not Detected in Effluent Gas During Many Analysis Periods

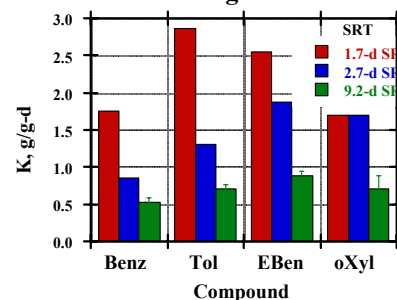
SRT days	Percent Analysis Periods Not Detected			
	Benz	Tol	Ethylb	o-Xyl
1.7	86	86	64	86
2.7	100	96	96	100
9.2	72	65	59	70

Gas Treatment Efficiency

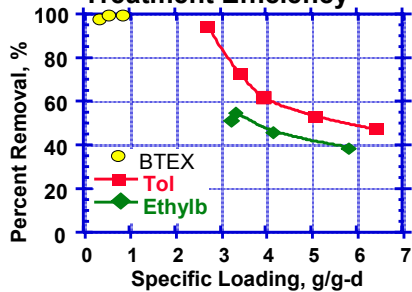
SRT days	Average Percent Removal			
	Benz	Tol	Ethylb	o-Xyl
1.7	99	99	99	99
2.7	99	99	98	98
9.2	98	97	97	98

* non detected effluent gas concentrations were set at the method detection limit (0.01 mg/L)

Biokinetics Changed with SRT



Effect of Specific Loading on Gas Treatment Efficiency



MODEL

Steady State Gas Treatment

$$C_{g-out} = C_{g-in} (\exp R) + H C_l (1 - \exp R)$$

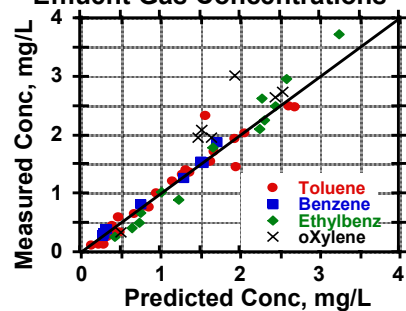
where:
$$R = \frac{-VI K_{la}}{Q_g H}$$

The VOC concentration in the reactor liquid is a function of the biodegradation kinetics:

$$\frac{(C_{g-in} - C_{g-out}) Q_g}{VI} = \frac{K C_l X}{K_s I + C_l}$$

I = inhibition constant
 $= 1 + C_l/b/K_s-b + C_l-c/K_s-c + \dots$

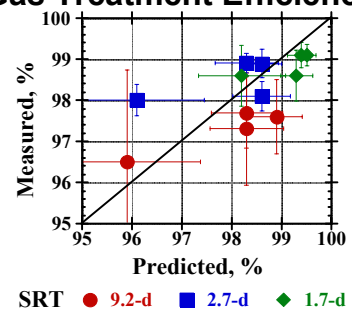
Measured versus Predicted Effluent Gas Concentrations



Steady-State Reactor Performance vs. Model Predictions

- Using the two equations, solved simultaneously for C_{g-out} and C_l
- Predicted liquid concentrations ranged from 0.01 to 0.03 mg/L for each BTEX compound, which agreed with measured values (< 0.05 mg/L detection limit)
- Predicted removal efficiencies agreed well with measured values

Gas Treatment Efficiency



When Cl is very low...

$$C_{g-out} = C_{g-in} (\exp R) + H C_l (1 - \exp R)$$

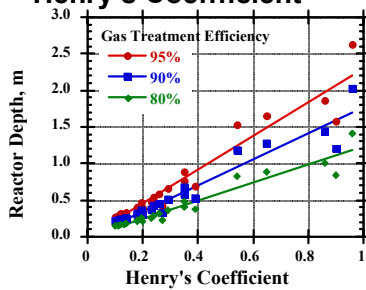
$$\text{where: } R = \frac{-V_l K_{la}}{Q_g H}$$

performance is controlled
by mass transfer

When Gas Treatment Efficiency is Not Biodegradation Limited, Design Reactor Depth Using:

$$\text{Depth} = \frac{-\ln(C_{g-out} / C_{g-in}) * Q_g / A * H}{K_{la}}$$

Required Reactor Depth Related to Henry's Coefficient



CONCLUSIONS

- High treatment efficiency (>98%) of a BTEX mixture was achieved
 - 40 cm depth
 - loading 15-17 mg/L-hr
 - SRTs 1.7, 2.7, 9.2 days

CONCLUSIONS

- A mechanistic model including mass transfer & biodegradation accurately predicted reactor performance
- The model can be used to aid full-scale design

CONCLUSIONS

- At all SRTs tested, liquid concentrations of BTEX were minimal, such that gas treatment was mass transfer limited
- For mass transfer limited designs, reactor depth is a function of H, K_{la}, Q_g/A