MCDB 4312/5312 Quantitative Optical Imaging

Prof Joel Kralj Fall 2017



Course Goals

- 1. Understand physics of light and optics
- 2. Explore contrast mechanisms for examine biological systems
- 3. Know strengths and limitations of fluorescent microscopy
- 4. Appreciate and understand new imaging techniques

Develop skills in Matlab to perform data analysis

- Class held on MWF 11:30 12:20 in JSCBB B331
- Grading:
 - 10% Classroom discussion/attendance
 - 40% Homeworks
 - 15% Midterm 1
 - 15% Midterm 2
 - 20% Final exam

- Midterms:
 - Sept 27th & Nov 1st
 - Midterms will cover only the immediately preceding section
- Final exam:
 - Scheduled by university registrar
 - Comprehensive across entire course
- Office hours
 - Open space in A-Wing on 3rd floor (Check in A321)
 - Mon 1-2PM, Thur 10-11 AM
 - Instructor Email: <u>Quant.Optics@Colorado.edu</u>
 - This is the only email on which I will respond to class questions

- Homeworks
 - Quantitative problems MUST HAVE work shown
 - Problem sets will consist of a theory portion and a Matlab portion
 - Homeworks must be turned in on time to receive credit
 - Email Matlab solutions to quant.optics@colorado.edu
- Discussion
 - Definitely ask questions during lectures. It is the best way of learning
 - Your questions will help me refine the course so you can learn more
 - Don't be upset if timing prohibits me from answering them

- Homework gestalt
 - I will ask open ended questions
 - You will have to access additional resources to find the answers
 - You are allowed to work with others in the class
 - Answers turned in must be your own
- Class pace
 - I will undoubtedly get excited, and move through material quickly
 - Please stop me and ask questions
 - Really, please stop me and ask if you have questions

- Appropriate learning environment
 - Cell phones off
 - Cheating is wrong
- Religious leave
 - Talk to me ASAP



Light, as seen by the ancients

- Transfer of energy across a vacuum
- Little tiny grains of sand, that moves from the sun to us
- Light connects two separated objects



Sunlight is composed of different colors

- Newton showed that white light is composed of unique colors, spread out by a prism
- Individual colors, if isolated, don't spread into other colors
- The rainbow can be remade back into white light
- White light is really the individual colors combined
- Light interacts with matter in very specific ways





Light travels in straight paths



Summing up (~1750):

- Light carries energy (across a vacuum)
- Light energy can interact with matter
- White light is composed of different colors
- Individual colors can not be further decomposed
- Light travels in straight lines

Pinhole camera – first attempt at imaging

- If there's a small hole, you can project an image onto another surface
- Trade off between light in and sharpness

Clear upside down (inverted) image with a small pinhole



Clear upside down (inverted) image with a small pinhole



Fuzzy out of focus image with a large pinhole

There's a problem

- Light acts like a wave
- Light undergoes diffraction, just like water or sound waves
- Light going through a small opening shows an oscillating pattern
- Light refracts, or bends, when travelling through a medium

Diffraction



Refraction



Double slit experiment

- Shows definitively that light is a wave
- Baseballs would behave like there is two independent slits
- Light forms a diffraction pattern if the slits are on the order of the wavelength





Light as a wave

- All waves have an amplitude and frequency, and they travel at a velocity dependent on the medium
- Energy of a wave is carried by amplitude, so brighter light has a bigger amplitude
- Color is carried by the frequency
- Maxwell's equations predict many properties of light, and that is a wave that travels at 3x10⁸ m/s
- Light is composed of electric and magnetic fields, and can be created by moving electrons
- Light displays many properties of waves, triumph of electromagnetic theory



Visible light

- Light is an oscillating electric and magnetic wave
- Color is defined by wavelength
- ROYGBIV
 - Red Orange Yellow Green Blue Indigo? Violet
- Spans the range from 400 nm – 700 nm



Electromagnetic spectrum



All is well, electricity and magnetism are solved, physics as a discipline starts winding down...



coefficient (cm⁻¹)

Absorption

Summing up (1905):

- Light is an electromagnetic wave
- Light carries energy (across a vacuum)
- Light energy can interact with matter
- White light is composed of different colors
- Colors are defined by wavelength (and frequency)
- Light waves travels in straight lines
- Light will diffract and refract

Yet another problem

- Light energy lands on a detector
- Detector will measure for some time, return a power (energy/second)
- Imagine steadily decreasing the light intensity of the source...
- Light comes in discrete bunches





Lumpiness is a fundamental property of light

- All sources are lumpy: lasers, sun, candles...
- Lumpiness is not dependent on color or complexity
- Image was scanned with increasing amounts of light
- All images are formed by adding up the number of lumps that land in a particular area





Fig. 1. Number of photons, 3×10^4 ; high-light luminance (foot-lamberts), 10^{-6} .

Fro. 4. Number of photons, 7.6×10⁵; high-light luminance (foot-lamberts), 2.5×10⁻⁴.





FIG. 5. Number of photons, 3.6×10°; high-light luminance (foot-lamberts), 1.2×10⁻⁹.





FIG. 3. Number of photons, 9.3×10⁴; high-light luminance (foot-lamberts), 3×10⁻⁶.

FIG. 6. Number of photons, 2.8×10³; high-light luminance (foot-lamberts), 9.5×10⁻⁶.

Wavelike property is localized

- Imagine a row of detectors in front of a very dim light source
- If this was a water wave, ie buoys in the ocean next to a rock that falls in, all the buoys would bob in synchrony
- Light triggers only individual detectors, with the same energy as if there was only a single detector



Light has both particle and wavelike features

- The double slit proved light is wavelike, but intensity proves it's lumpy
- Einstein proposed that light comes in discrete packets of energy
- Each packet energy is set by the frequency, and all packets of the same color have the same energy
- Each of these particles are called photons

$$E = hf$$
$$= hc/\lambda$$



h = Planck's constant = $6.62 \times 10^{-34} \text{ m}^2\text{kg/s}$

Quantum weirdness

- A double slit experiment, even with single photons still produces a diffraction pattern
- Monitoring either slit for the passage of the photon will destroy diffraction pattern
- The act of measurement changes photons from waves to particles





Matlab tutorial

- <u>https://oit.colorado.edu/software-hardware/software-downloads-and-licensing/matlab</u>
- Accessing Matlab from the desktop
- Screen layout