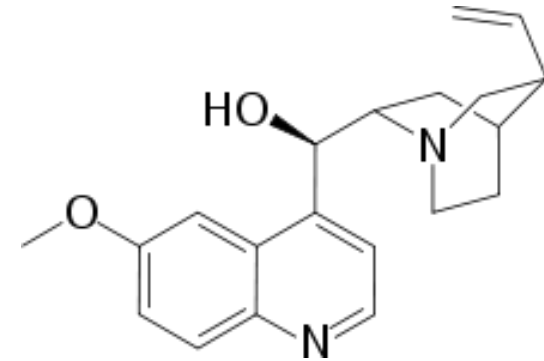


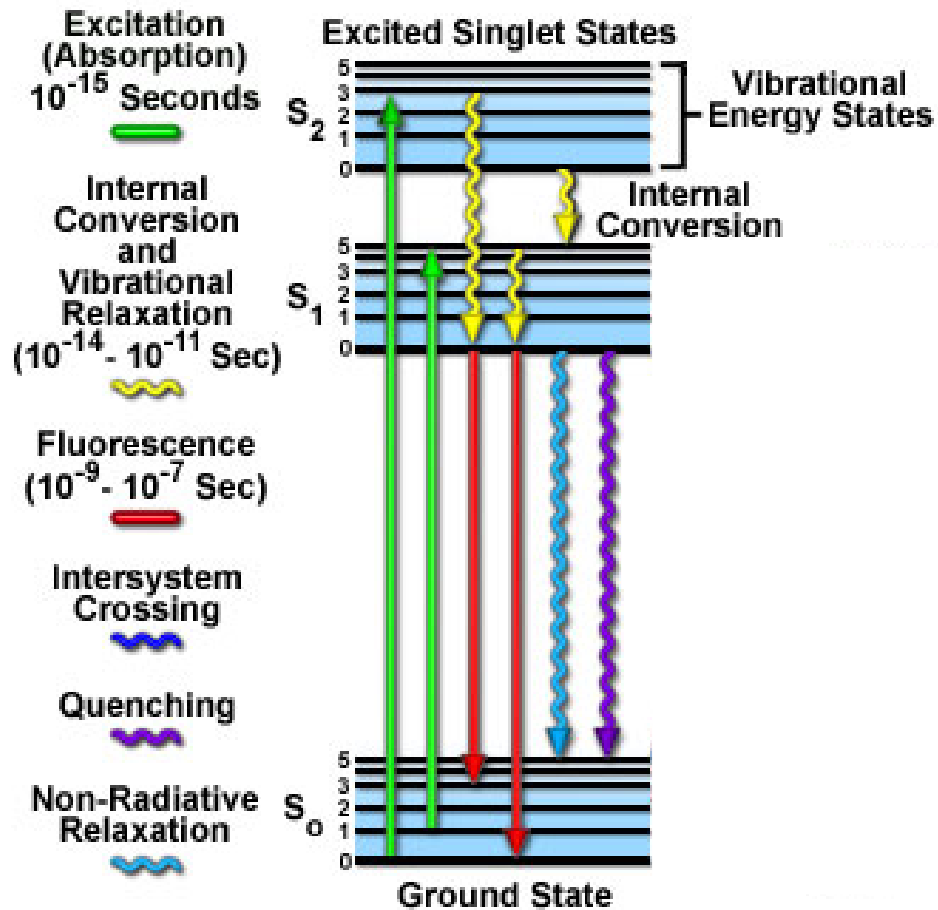
More on fluorescence

- Last class
 - Fluorescence
 - Absorption emission
 - Jablonski diagrams
- This class
 - More on fluorescence
 - Common fluorophores

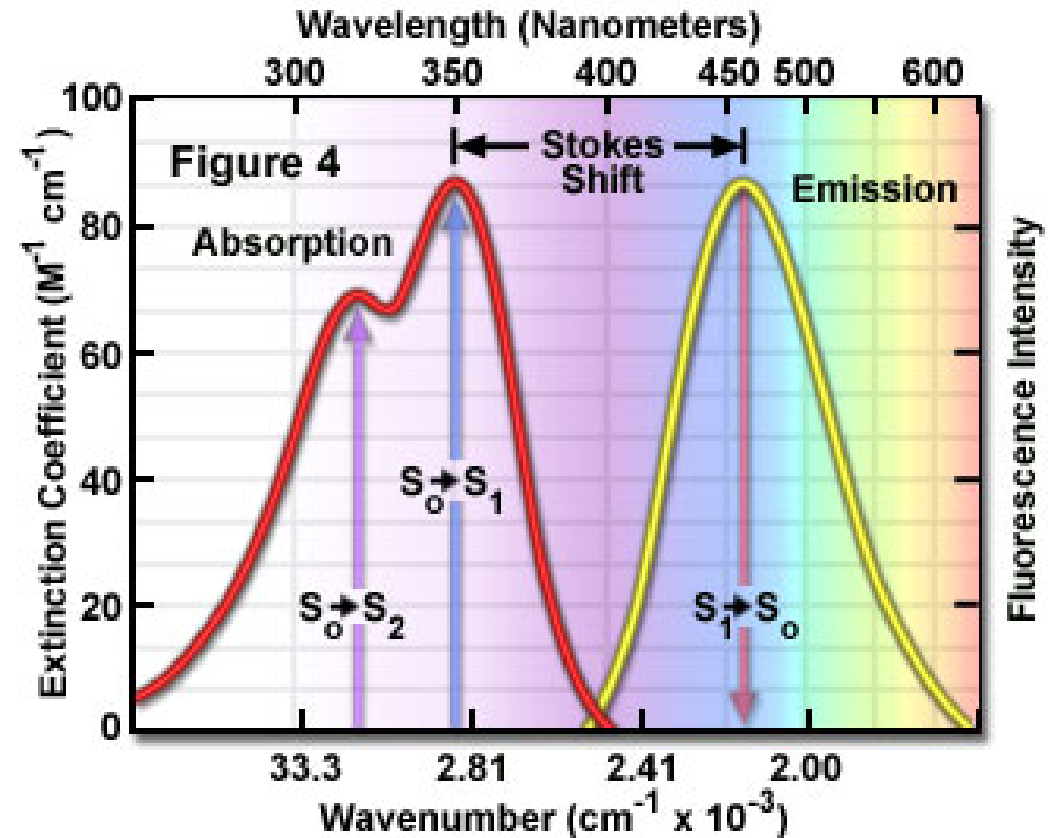
Jablonski diagrams to spectra



Jablonski Energy Diagram

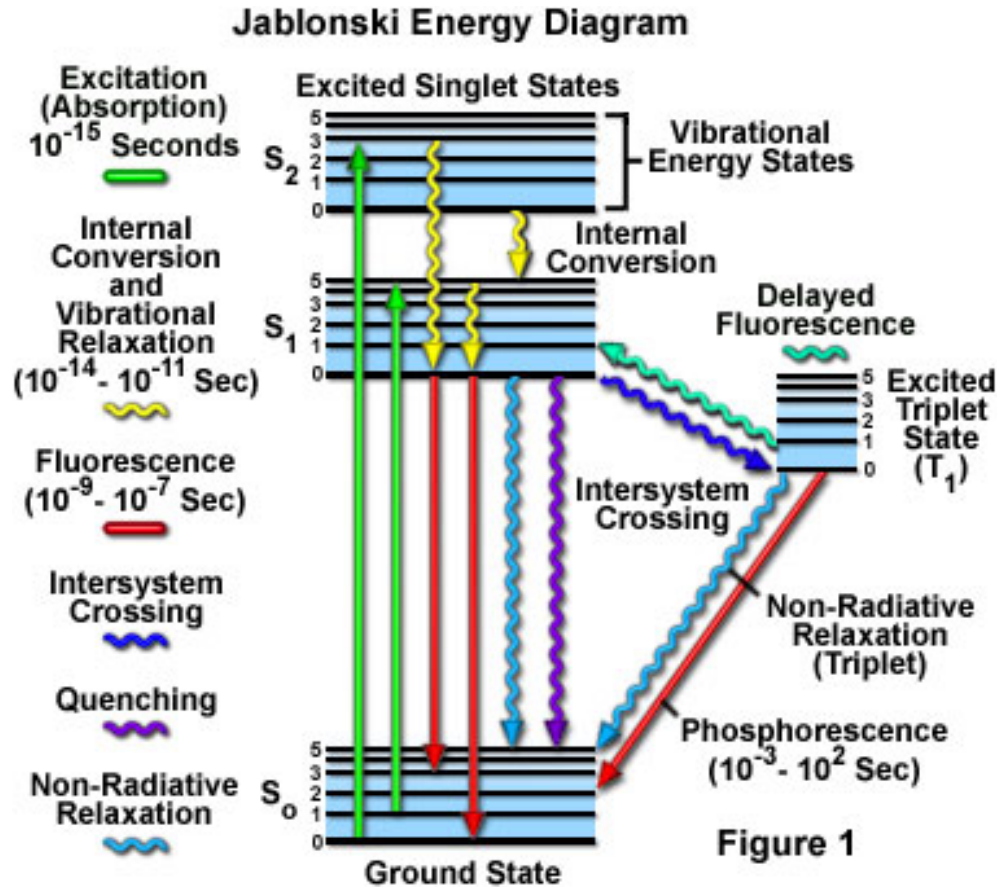


Quinine Absorption and Emission Spectra



Properties of fluorophores

- Excitation max
- Emission max
- Spectrum breadth
- Molar extinction coefficient
- Quantum yield
- Photostability
- Photons per molecule
- Solubility

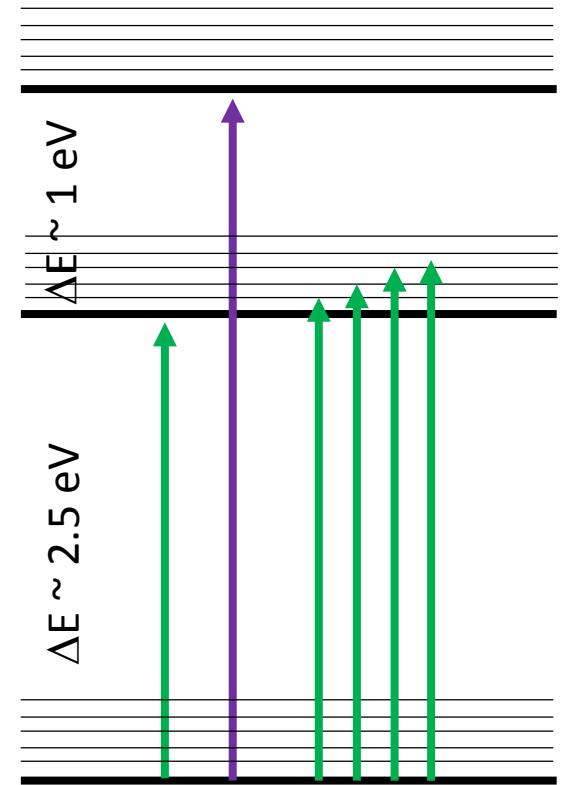
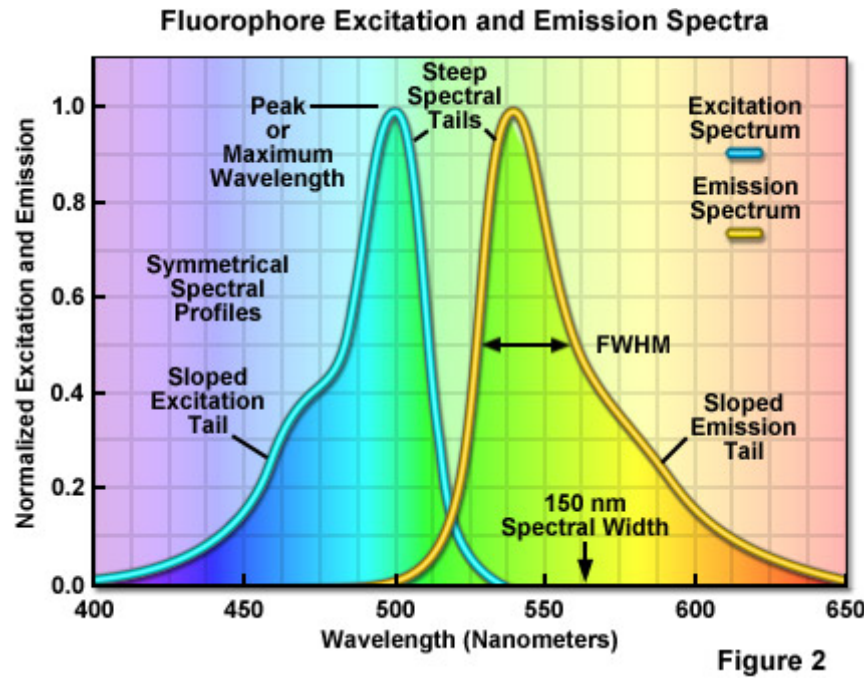


$$QY = \frac{\#Fl \text{ photons}}{\#Abs \text{ photons}}$$

$$QY = \frac{k_f}{k_f + k_{NR}}$$

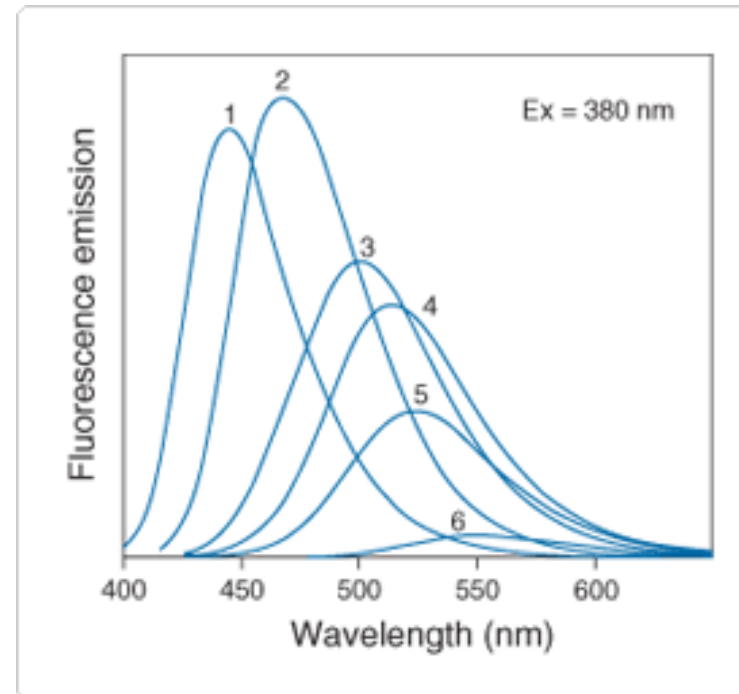
Spectra of fluorophores

- Absorption – the ability to absorb photons at a given wavelength
- Excitation – ability to excite fluorescence at a given wavelength
- Emission – ability to emit photons at a given wavelength



Fluorescence emission is determined by local environment

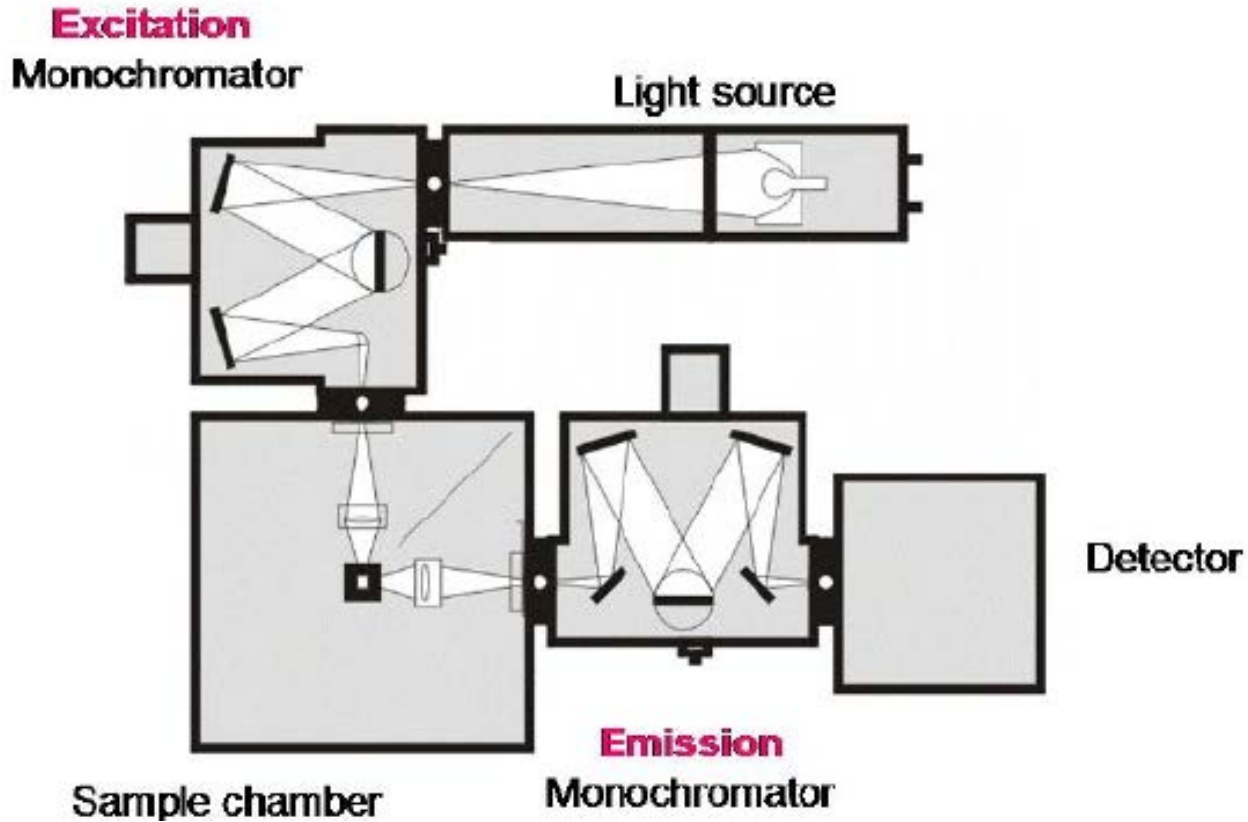
- Temperature
- pH
- Oxygen content
- Solvent
- Concentration of fluorophore



1) toluene, 2) chloroform, 3) acetonitrile, 4) ethanol, 5) methanol and 6) water

$$\text{Brightness} = \epsilon * QY/1000$$

Measuring fluorescence



Start with known concentration of fluorophore

Run a quick excitation and emission scan

Compare to known fluorophore

$$\frac{QY_1 abs_1}{em_1} = \frac{QY_2 abs_2}{em_2}$$

Abs = absorbance at specific wavelength

Em = integrated emission

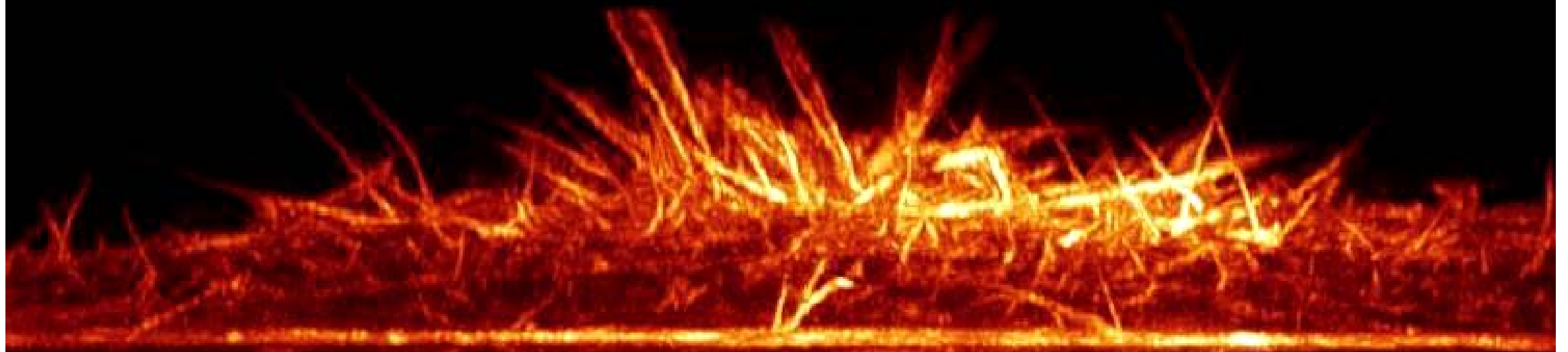
QY = quantum yield

Fluorescence imaging

- Why do we use it so much?
- Color separation gives it power
- We can use filters to look for ONLY the emitted light
- Low background -> High contrast
- Works in live cells
- Small sizes
- We are good at dealing with the visible spectrum

$$C = \left[\frac{I_b - I_s}{I_b} \right] * 100\%$$

Lattice SIM 3 phase

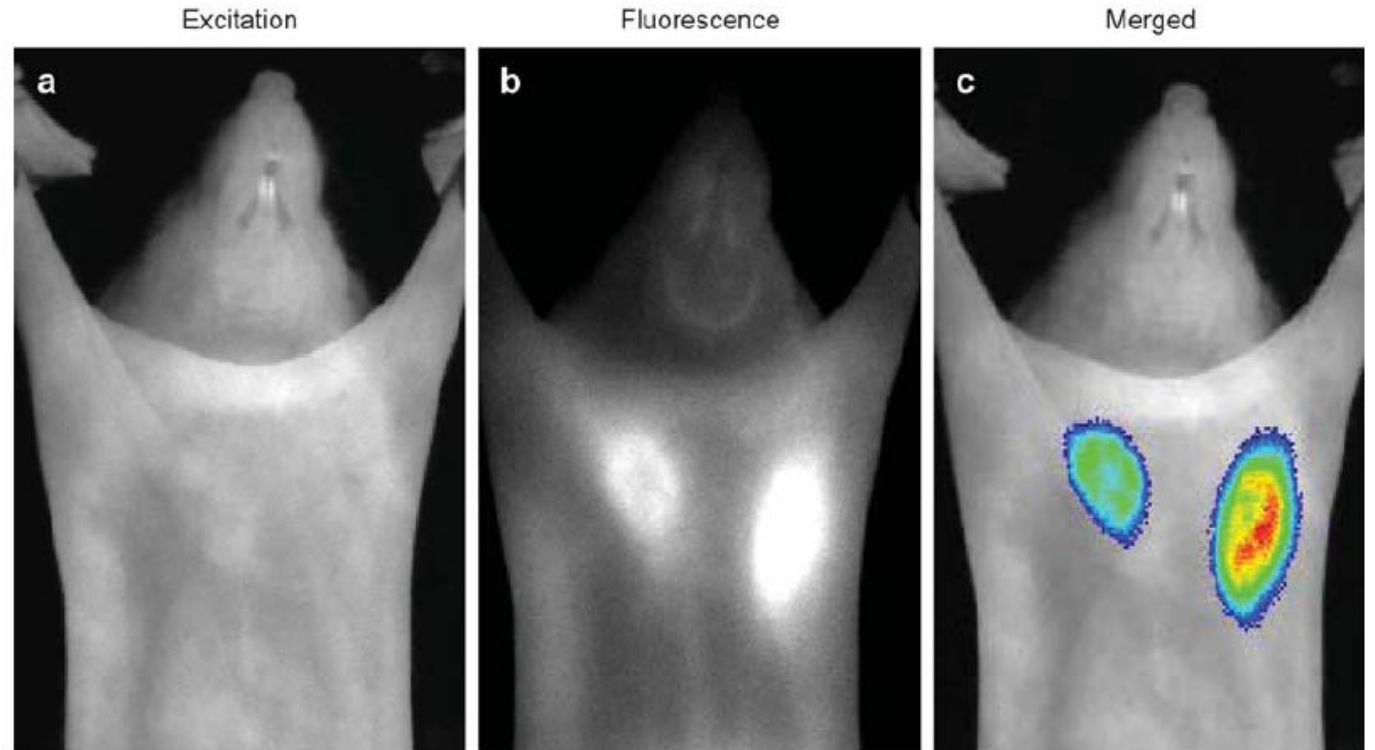


Stack = 0
Time = 0.00 mins

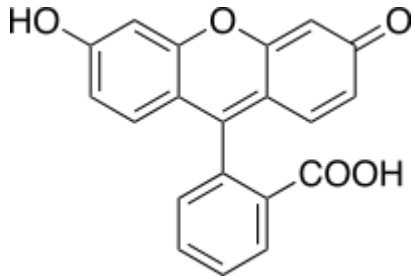
HeLa cell mEmerald - Lifeact

Applications of fluorescence in biology

- Intracellular distribution
- Intracellular dynamics
- Protein interactions
- Intracellular sensing
- Organelle marking/status
- Enzyme reactions
- Cell physiology
- Neuronal tracing
- Cell fate measurements
- Cell tracking
- Cell cycle



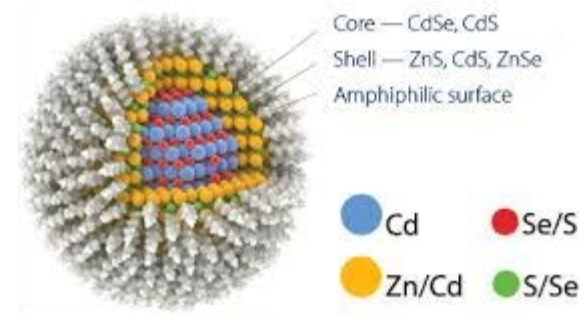
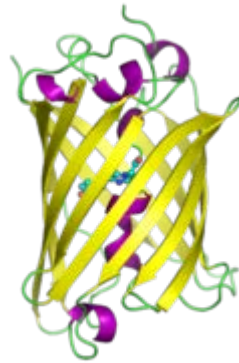
Types of fluorophores



Dyes



Proteins

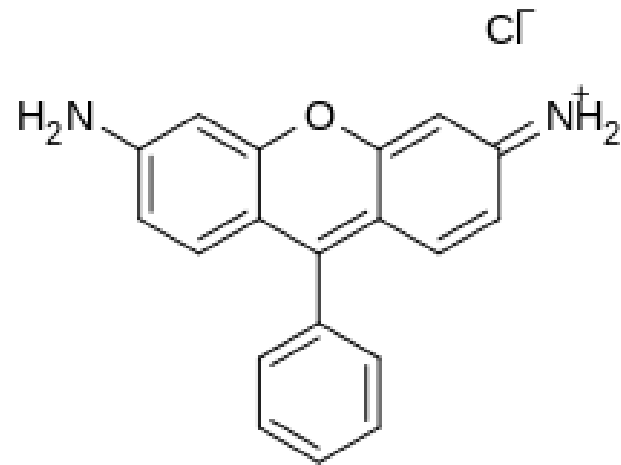


Quantum dots



Dyes

- Must have extended polar electron chain
- Extended electron density



Cheap, common

	Excitation	Emission
DAPI	358	461
FITC	495	519
TRITC	558	576
Texas red	589	615
Cy3	550	570
Cy5	650	670
Cy7	743	767

Metrics

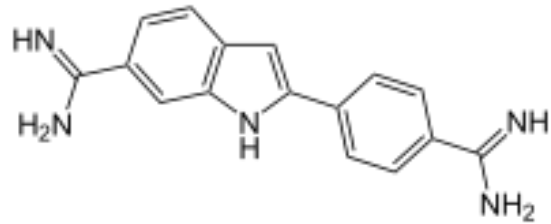
1. Excitation/emission
2. Quantum yield
3. Brightness
4. Photostability
5. Toxicity
6. Solubility
7. Permeability

Alexa Fluor series

	Color	Absorb	Emit	MM	ϵ	Quantum Yield ^[6]
Alexa Fluor 350	blue	346	442	410	19,000	-
- 405	violet	401	421	1028	35,000	-
- 430	green	434	541	702	15,000	-
- 488	cyan-green	495	519	643	73,000	0.92
- 500	green	502	525	700	71,000	-
- 514	green	517	542	714	80,000	-
- 532	green	532	554	721	81,000	0.61
- 546	yellow	556	573	1079	112,000	0.79
- 555	yellow-green	555	565~1250		155,000	0.1
- 568	orange	578	603	792	88,000	0.69
- 594	orange-red	590	617	820	92,000	0.66
- 610	red	612	628	1172	144,000	-
- 633	Far-red	632	647~1200		159,000	-
- 635	Far-red	633	647-		140,000	-
- 647	Far-red	650	665 1155.06[7]		270,000	0.33
- 660	Near-IR	663	690~1100		132,000	0.37
- 680	Near-IR	679	702~1150		183,000	0.36
- 700	Near-IR	702	723~1400		205,000	0.25
- 750	Near-IR	749	775~1300		290,000	0.12
- 790	Near-IR	782	805-		260,000	-

DAPI – nuclear stain

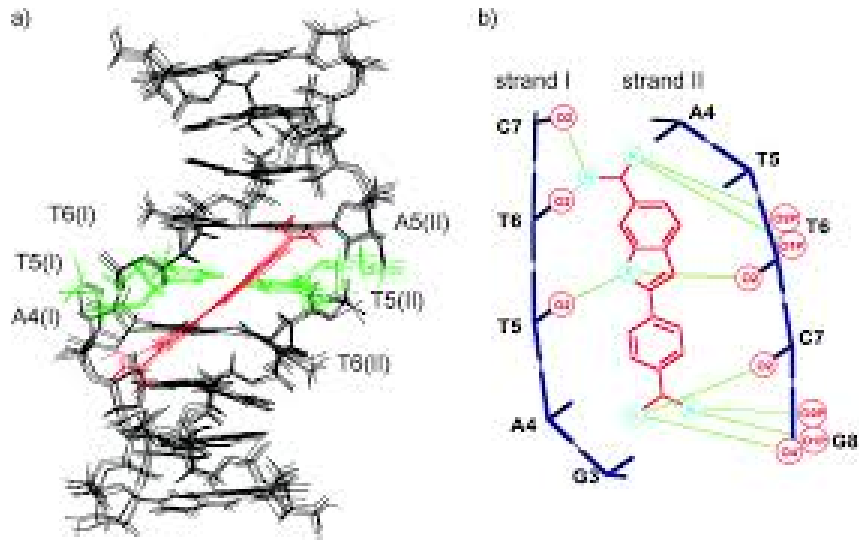
(4',6-diamidino-2-phenylindole)



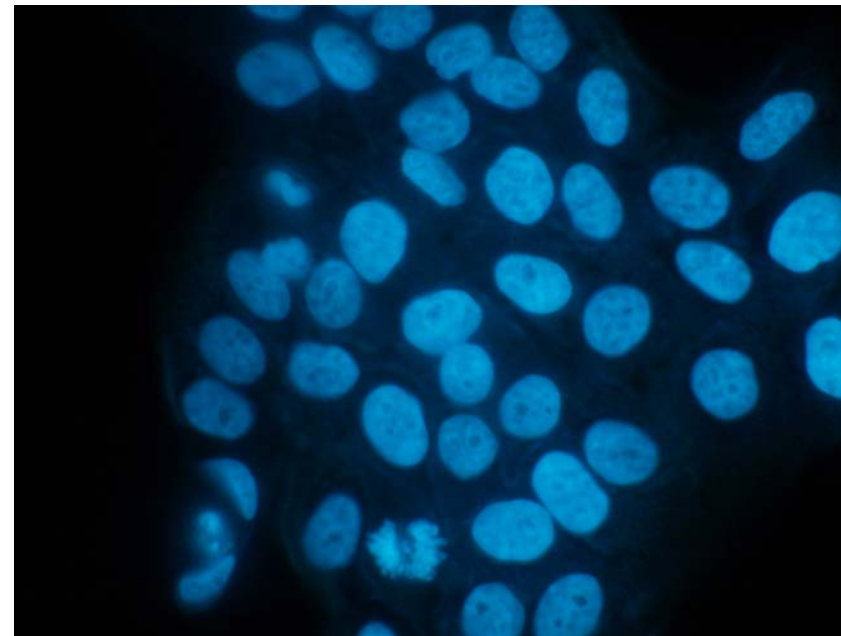
Binds in AT rich regions of DNA
Increases fluorescence upon nucleic acid binding

Excitation/Emission – UV/blue

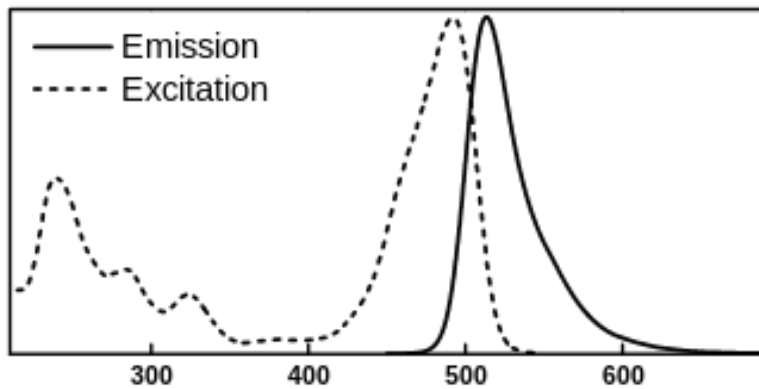
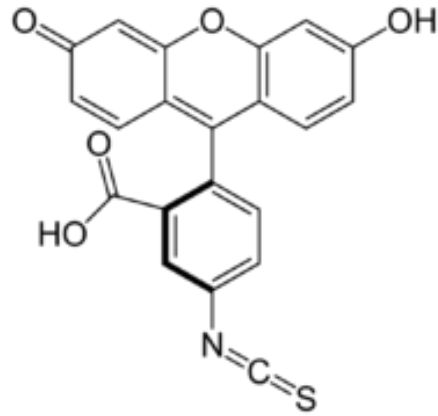
Very common to label nuclei in fixed cells. Can be used in live cells, but toxic



Laser line = 405 nm



FITC - fluorescein



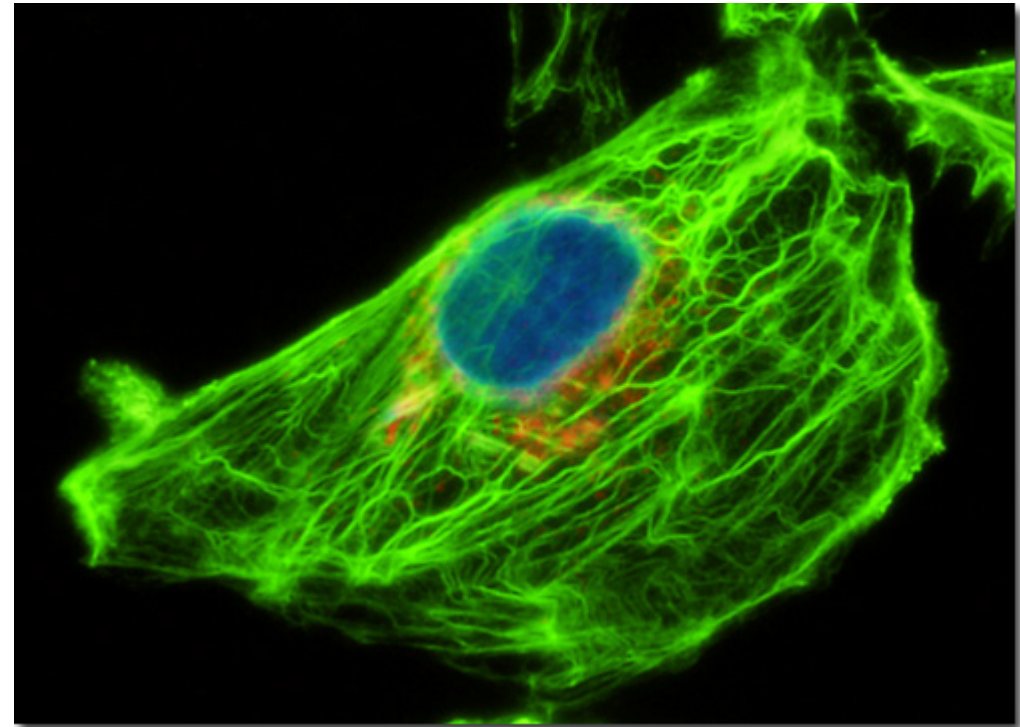
Laser line = 488 nm

Excitation/Emission – 495/515 nm

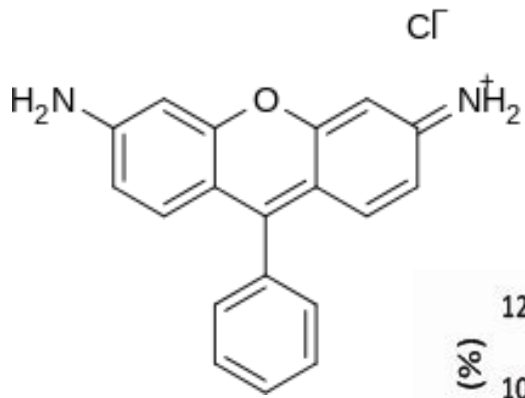
Very common fluorescent dye – super cheap. Similar spectrum to GFP.

Highly sensitive to pH, not very photostable

Common in immunofluorescence



TRITC - tetramethylrhodamine

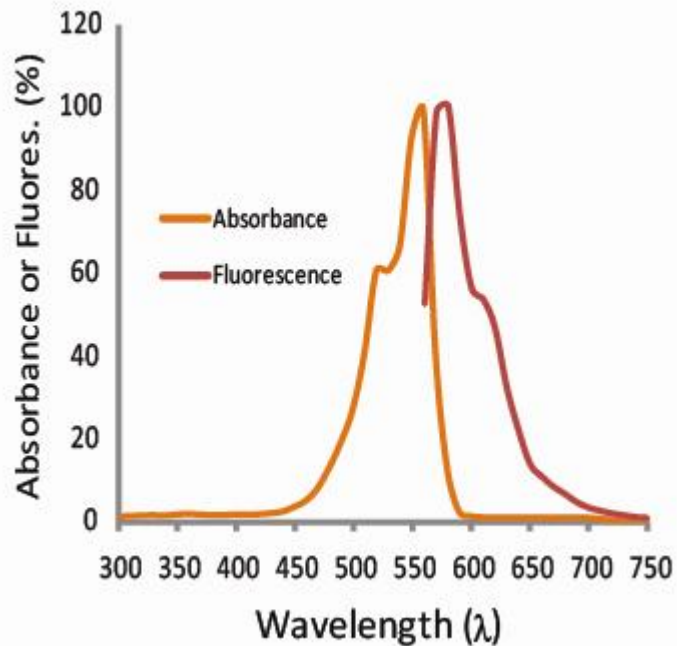


Excitation/Emission – 550/579 nm

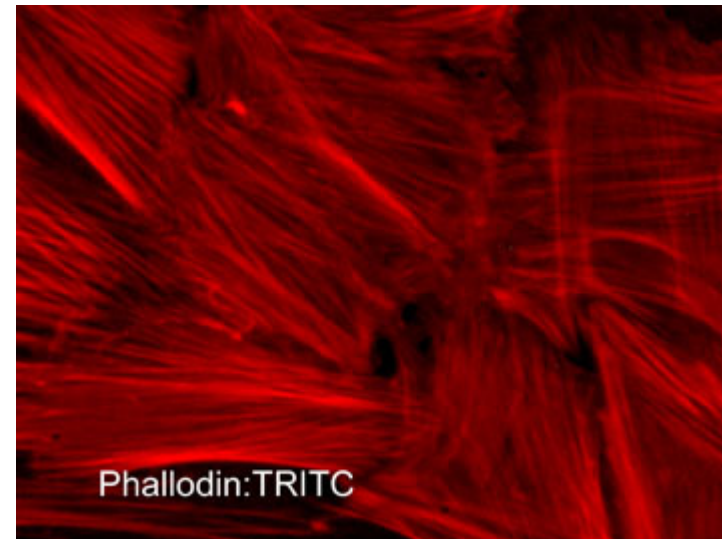
Very common fluorescent dye – super cheap. Similar spectrum to mCherry.

Not very photostable

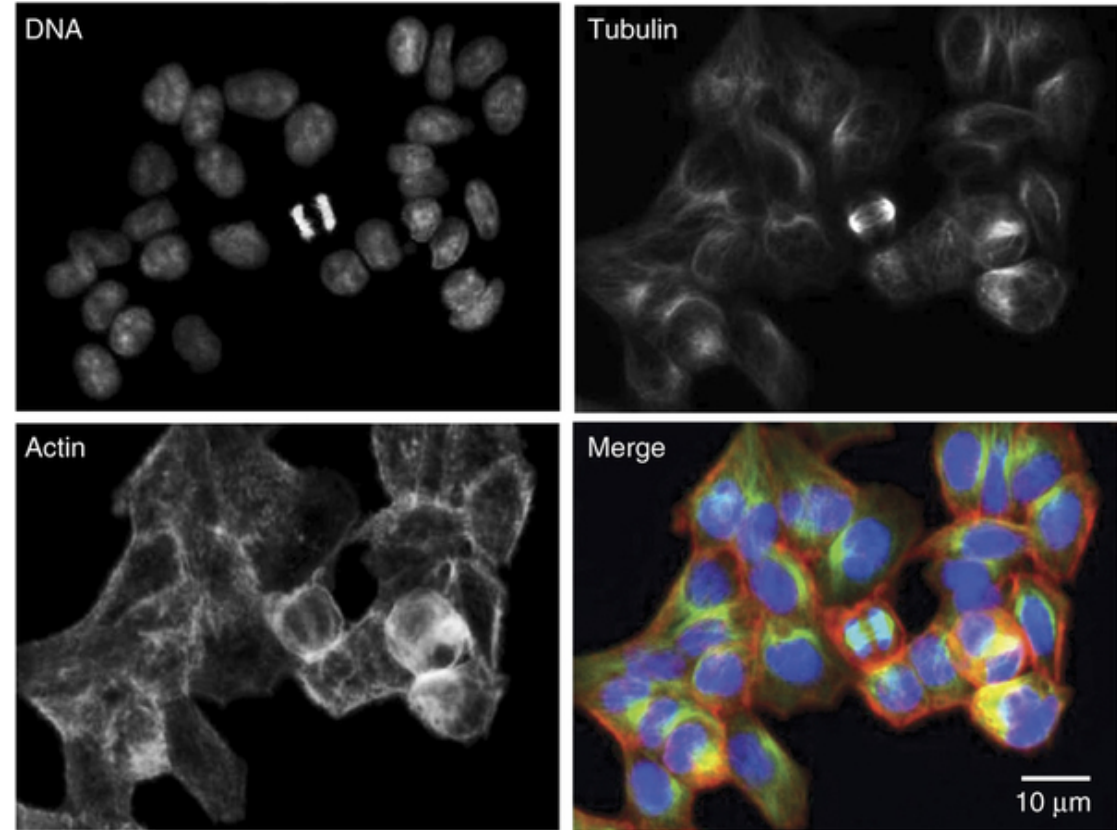
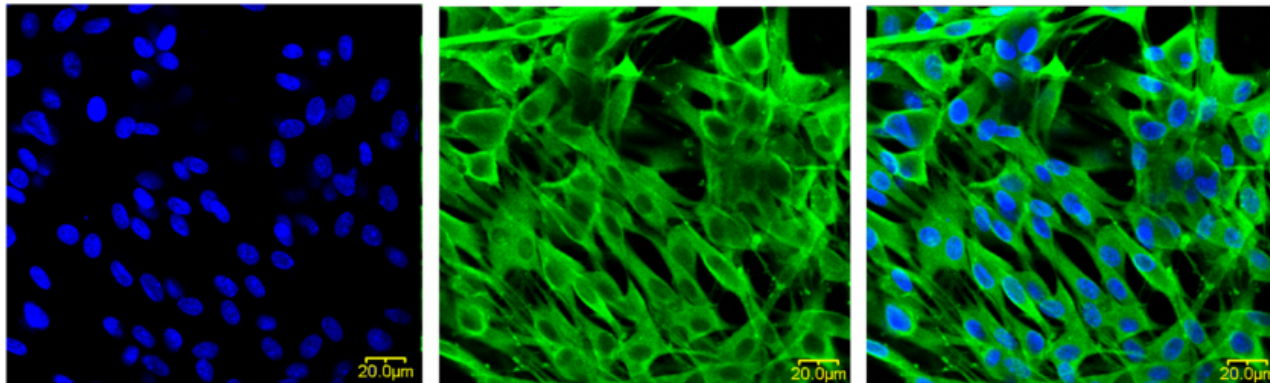
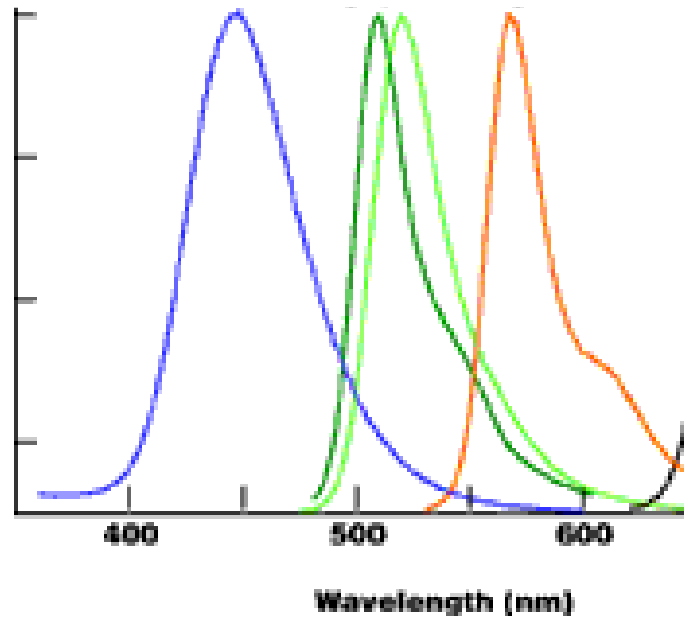
Common in immunofluorescence



Laser line = 532 nm, 561 nm

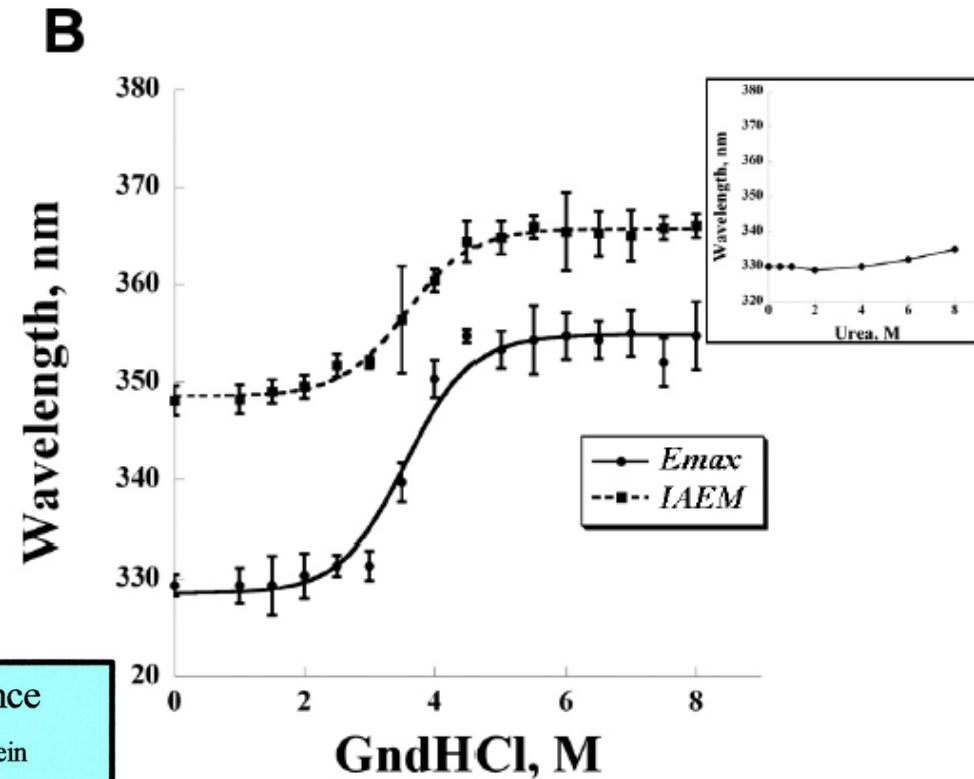
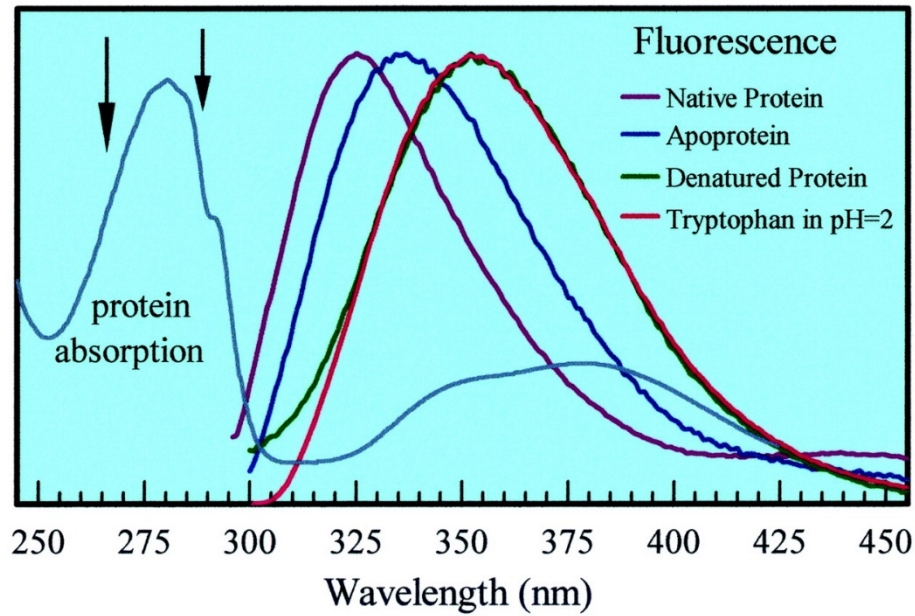
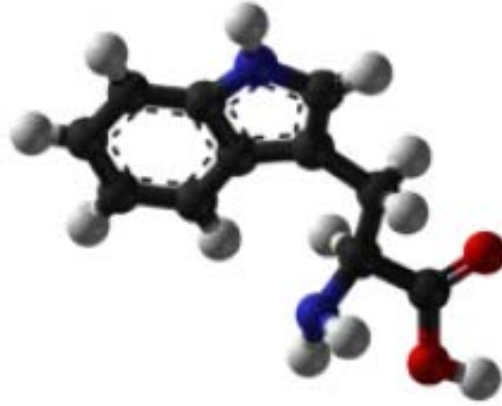


DAPI-FITC-TRITC are a common set



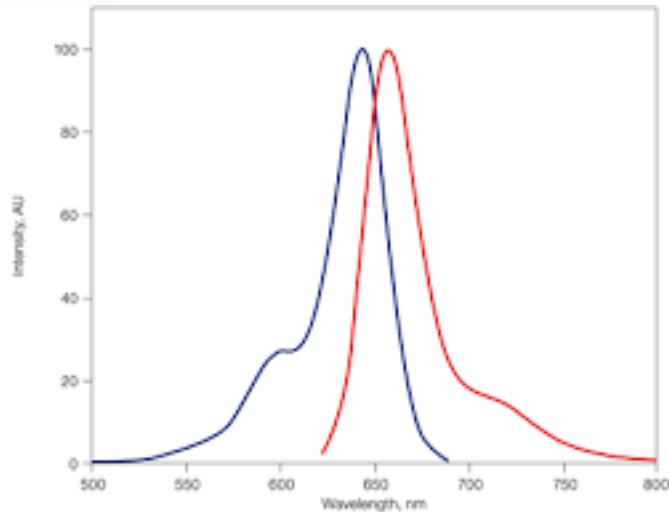
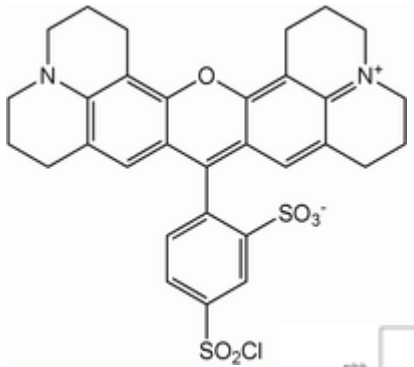
Tryptophan fluorescence

- Aromatic amino acid
- Emits at 330 nm in hydrophobic environment
- Emits at 365 nm in polar environments
- Can be used to track protein folding



Texas Red

Sulforhodamine 101



Laser line = 594 nm

Excitation/Emission – 590\610 nm

Common fluorescent dye
Somewhat photostable

Common in immunofluorescence

Used to extend range of visible light,
but can't be used with TRITC

Nikon Texas Red HYQ Yellow Excitation Bandpass Filter Set

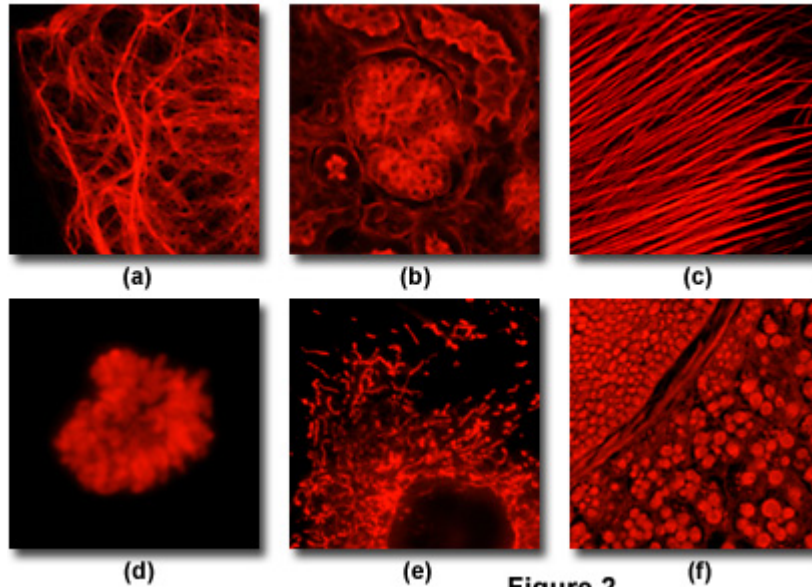


Figure 2

Cyanine variants

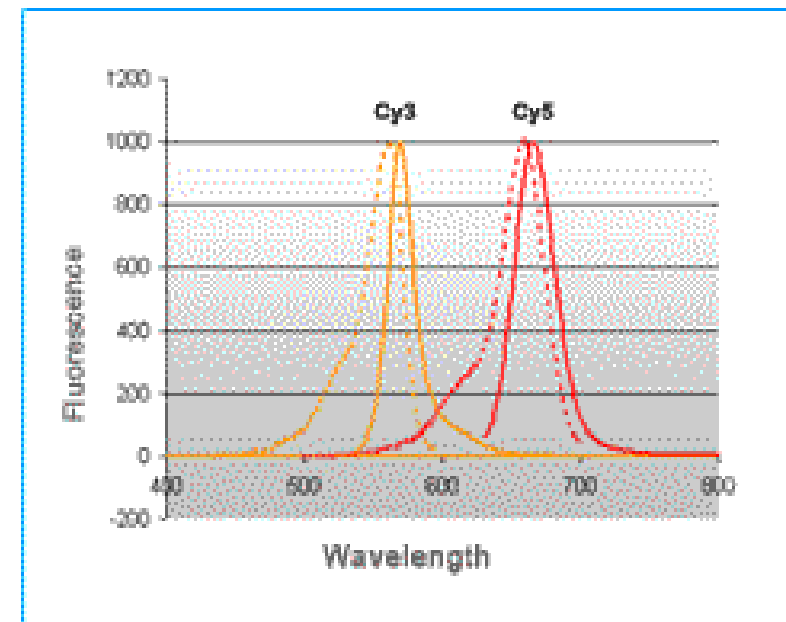
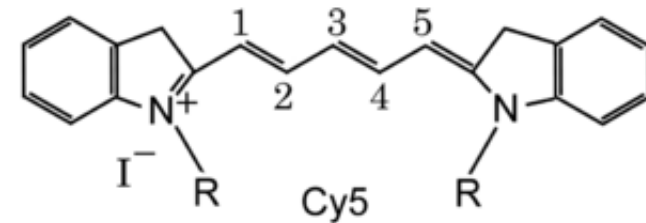
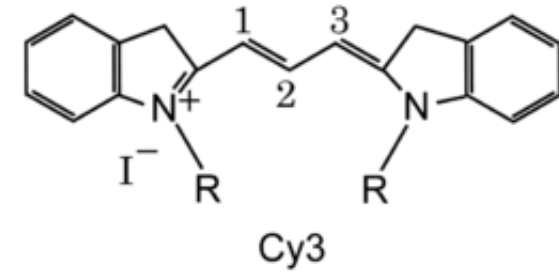
Probe	Ex (nm)	Em (nm)	MW	Quantum yield
Cy2	489	506	714	QY 0.12
Cy3	(512);550	570;(615)	767	QY 0.15 [5]
Cy3B	558	572;(620)	658	QY 0.67
Cy3.5	581	594;(640)	1102	QY 0.15
Cy5	(625);650	670	792	QY 0.27 [5]
Cy5.5	675	694	1128	QY 0.28 [6]
Cy7	743	767	818	QY 0.28

Cy3 is a newer TRITC equivalent

Cy3 and Cy5 are by far the most common

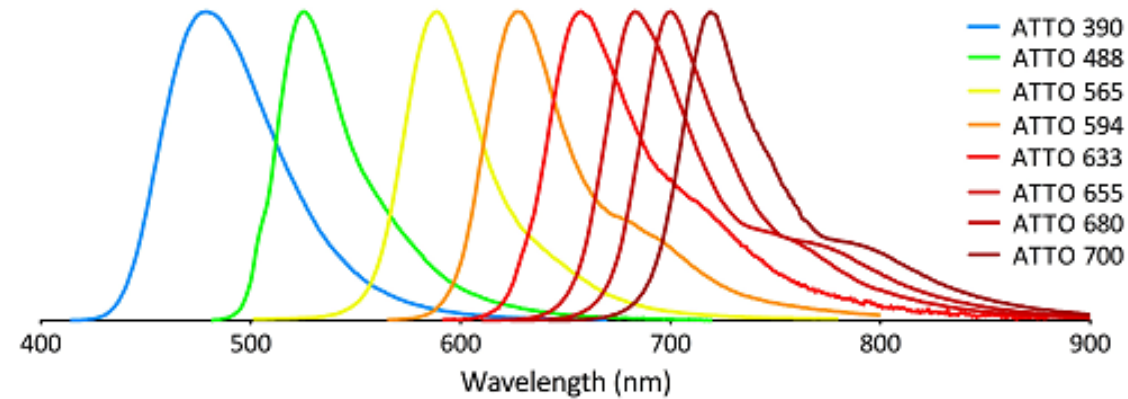
Cy5 can add a 4th color:

DAPI, FITC, TRITC (or Cy3), Cy5



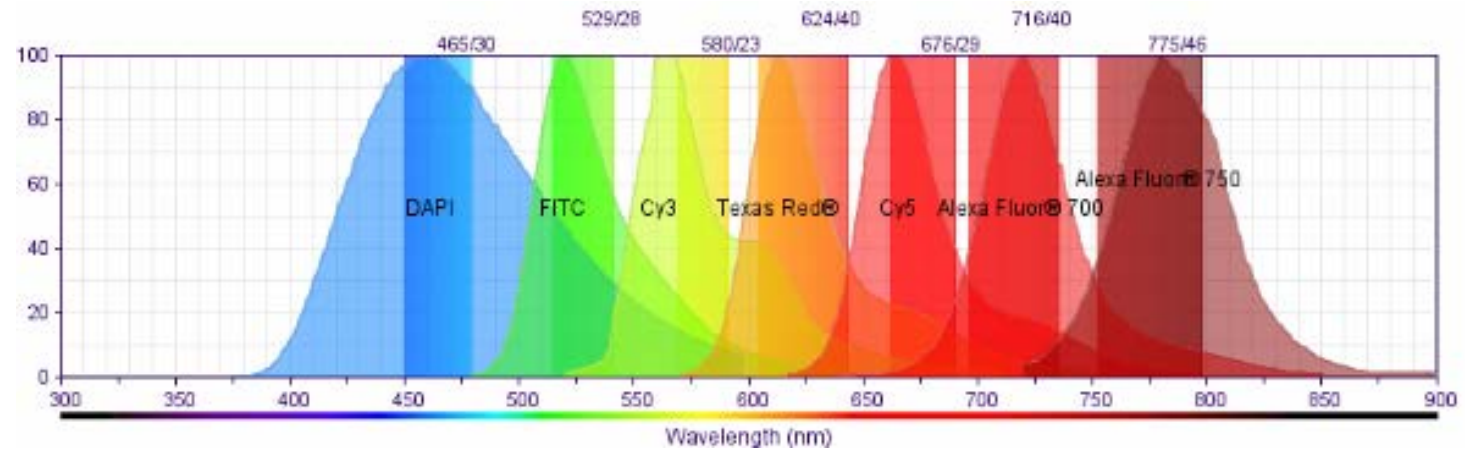
Modern variants

- Cyanine dyes
- Alexa fluor series
- Atto series
- Dylight series



More money =

1. Higher brightness
2. Photostability
3. Water solubility
4. Broad range of colors



And on to Matlab...