

Constraining U.S. ammonia emissions using TES remote sensing observations and the GEOS-Chem adjoint model

L. Zhu,¹ D. K. Henze,¹ K. E. Cady-Pereira,² M. W. Shephard,^{3,4} M. Luo,⁵
R. W. Pinder,⁶ J. O. Bash,⁶ and G.-R. Jeong^{1,6}

Received 2 August 2012; revised 25 December 2012; accepted 3 January 2013.

[1] Ammonia (NH₃) has significant impacts on biodiversity, eutrophication, and acidification. Widespread uncertainty in the magnitude and seasonality of NH₃ emissions hinders efforts to address these issues. In this work, we constrain U.S. NH₃ sources using observations from the TES satellite instrument with the GEOS-Chem model and its adjoint. The inversion framework is first validated using simulated observations. We then assimilate TES observations for April, July, and October of 2006 through 2009. The adjoint-based inversion allows emissions to be adjusted heterogeneously; they are found to increase in California throughout the year, increase in different regions of the West depending upon season, and exhibit smaller increases and occasional decreases in the Eastern U.S. Evaluations of the inversion using independent surface measurements show reduced model underestimates of surface NH₃ and wet deposited NH_x in April and October; however, the constrained simulation in July leads to overestimates of these quantities, while TES observations are still under predicted. Modeled sulfate and nitrate aerosols concentrations do not change significantly, and persistent nitrate overestimation is noted, consistent with previous studies. Overall, while satellite-based constraints on NH₃ emissions improve model simulations in several aspects, additional assessment at higher horizontal resolution of spatial sampling bias, nitric acid formation, and diurnal variability and bi-directionality of NH₃ sources may be necessary to enhance year-round model performance across the full range of gas and aerosol evaluations.

Citation: Zhu, L., D. K. Henze, K. E. Cady-Pereira, M. W. Shephard, M. Luo, R. W. Pinder, J. O. Bash, and G.-R. Jeong (2013), Constraining U.S. ammonia emissions using TES remote sensing observations and the GEOS-Chem adjoint model, *J. Geophys. Res. Atmos.*, 118, doi:10.1002/jgrd.50166.

1. Introduction

[2] Emissions of ammonia (NH₃) from anthropogenic sources pose several environmental concerns. Ammonia affects air quality and climate through its role in the mass, composition, and physical properties of tropospheric aerosol. Ammonium nitrate and ammonium sulfate make up

a substantial fraction of atmospheric fine particulate matter (PM_{2.5}), exposure to which has been statistically associated with inhibited lung development, cardiovascular diseases, and premature mortality [Pope *et al.*, 2002; Schwartz *et al.*, 2002; Reiss *et al.*, 2007]. These fine particulates (PM_{2.5}) also contribute to haze and thus impact visibility. Further, when deposited in excess, reactive nitrogen, including ammonia, can cause detrimental nutrient imbalances to sensitive ecosystems [Rodhe *et al.*, 2002; Rabalais, 2002].

[3] Despite the recognized importance of NH₃ emissions in the U.S. [Aneja *et al.*, 2008], knowledge of their magnitude is severely limited; NH₃ emissions are primarily from agricultural sources whose strengths are difficult to characterize. Uncertainty in NH₃ undermines efforts to understand historical and present levels of PM_{2.5} [Yu *et al.*, 2005; Nowak *et al.*, 2006; Zhang *et al.*, 2008; Wu *et al.*, 2008; Stephen and Aneja, 2008; Beusen *et al.*, 2008; Simon *et al.*, 2008; Henze *et al.*, 2009] and hinders estimates of the response of PM_{2.5} to control measures because of the key role that NH₃ plays in governing the balance of inorganic fine particulate species [Dennis *et al.*, 2008]. Model estimates of inorganic PM_{2.5} have been compared to surface measurements [Park *et al.*, 2004, 2006; Liao *et al.*, 2007; Henze *et al.*, 2009; Pye *et al.*, 2009;

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¹Department of Mechanical Engineering, University of Colorado, Boulder, Colorado, USA.

²Atmospheric and Environmental Research, Inc., Lexington, Massachusetts, USA.

³Atmospheric and Climate Application, Inc., East Gwillimbury, Ontario, Canada.

⁴Environment Canada, Toronto, Ontario, Canada.

⁵Jet Propulsion Laboratory, California Institute of Technology, Pasadena, California, USA.

⁶US Environmental Protection Agency, Research Triangle Park, North Carolina, USA.

Corresponding author: D. K. Henze, Department of Mechanical Engineering, University of Colorado, Boulder, CO, USA. (daven.henze@colorado.edu)

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2169-897X/13/10.1002/jgrd.50166

Heald *et al.*, 2012] and measurements from aircraft campaigns [Heald *et al.*, 2005, 2006]; NH₃ emissions are frequently indicated to be a likely cause of discrepancies. On a larger scale, NH₃ emissions rates are a critical source of uncertainty in global budgets of the atmospheric transport and deposition of reactive nitrogen [Sutton *et al.*, 2007; Galloway *et al.*, 2008; Schlesinger, 2009].

[4] These are several reasons for the persistence of uncertainties in NH₃ inventories. Characterizing NH₃ sources from the bottom up requires spatially and temporally resolved data such as detailed farming practices and intensity. These data are rarely available nationally as direct measurements of NH₃ emissions at such scales are prohibitive owing to cost. Therefore, top-down approaches have become an attractive option for providing additional constraints. While direct observations of gas-phase NH₃ do exist in select locations, observations of other chemically related species are much more prevalent. Furthermore, NH₃ can rapidly partition to form aerosol ammonium (NH₄⁺) which can limit the utility of gas-phase observations alone.

[5] Consequently, owing to the paucity of direct observations of NH₃ and the difficulty of constraining the NH_x (=NH₃ + NH₄⁺) system, measurements of species that are regulated by the amount of available NH₃ have been looked to for constraints on estimates of NH₃ emissions. The National Emissions Inventory (NEI) for NH₃ is coarsely constrained by top-down estimates from the inverse modeling studies of Gilliland *et al.* [2003, 2006]. Measurements of wet deposited NH_x were used as constraints, because wet deposited NH_x estimates depend less than NH₃ on model sensitivity to aerosol partitioning. A drawback to this approach is the sensitivity to the precipitation fields in the meteorological re-analysis data driving the chemical transport model and to the parameterization of NH_x wet scavenging, both of which are aspects that are difficult to model accurately and hinder the inversion during some seasons. Taking an alternate approach, Henze *et al.* [2009] used surface measurements of SO₄²⁻ and NO₃⁻ from the IMPROVE network to constrain the amount of NH₃ partitioned into the aerosol phase as NH₄⁺ (which is strongly coupled to SO₄²⁻ and NO₃⁻). In this way, aerosol-phase observations were used to constrain NH₃ concentrations and, hence, NH₃ emissions. This approach, however, is sensitive to model bias in HNO₃, which may be significant [Zhang *et al.*, 2012; Heald *et al.*, 2012].

[6] Despite these recent efforts, comparisons of inverse modeling results to the bottom-up NH₃ inventory of Pinder *et al.* [2006] show that considerable disagreements remain in the spatial and seasonal distribution of NH₃ emissions throughout the U.S. [Henze *et al.*, 2009]; at odds are estimates of the relative magnitude of spring versus summer emissions. A limiting factor in reconciling these differences is infrequent and sparse in situ observations, even for the aerosol-phase measurements, and a shortage of direct constraints on gas-phase NH₃. Without understanding the NH_x system as a whole, and without tools to link observations of these species over the continent to emissions, studies of NH₃ or NH₄⁺ alone may suffer in terms of utility for constraining emissions inventories at a national scale [Pinder *et al.*, 2006].

[7] The detection of boundary layer ammonia from space [Beer *et al.*, 2008; Clarisse *et al.*, 2009, 2010; Shephard *et al.*, 2011] provides a new opportunity for reducing persistent uncertainties in our understanding of the distribution and impacts of atmospheric ammonia. Initial comparisons to global model NH₃ distributions indicate that NH₃ sources may be widely underestimated [Clarisse *et al.*, 2009; Shephard *et al.*, 2011]. Pinder *et al.* [2011] have verified the utility of such measurements for tracking observed spatial and temporal trends in surface level NH₃ concentrations. Subsequent studies indicate that underestimates of NH₃ sources exist in California and throughout the U.S. in the spring [Nowak *et al.*, 2012; Walker *et al.*, 2012; Heald *et al.*, 2012]. Therefore, we consider here how inverse modeling with assimilation of satellite observations of NH₃ can be used to further provide constraints on NH₃ sources. Section 2 describes the models and inverse methodology used in this study. We then present details of the remote sensing observations (Section 3), followed by inverse modeling tests using simulated observations (Section 4) and real observations (Section 5). Finally, we evaluate the model results by comparing them to independent data sets omitted from the inversion (Section 6) and present our conclusions (Section 7).

2. Methods

2.1. GEOS-Chem

[8] GEOS-Chem is a chemical transport model driven with assimilated meteorology from the Goddard Earth Observing System (GEOS) of the NASA Global Modeling and Assimilation Office [Bey *et al.*, 2001]. The tropospheric oxidant chemistry simulation in GEOS-Chem includes a detailed ozone-NO_x-hydrocarbon chemical mechanism of 80 species and over 300 reactions [Bey *et al.*, 2001]. GEOS-Chem includes an online secondary inorganic aerosol simulation introduced and described in full by Park *et al.* [2004]. Global anthropogenic sources and natural sources of NH₃ are from the 1990 GEIA inventory [Bouwman *et al.*, 1997]. Over the U.S., anthropogenic NH₃ emissions are taken from the 2005 U.S. National Emissions Inventory (NEI), with seasonality as described in Park *et al.* [2004], and the CAC inventory for Canada [van Donkelaar *et al.*, 2008]. Biomass burning emissions are from van der Werf *et al.* [2006], and biofuel emissions are from Yevich and Logan [2003]. The total U.S. and global NH₃ emissions both before and after the optimization are shown in Table 1.

Table 1. Total U.S. and Global NH₃ Emissions Before and After Optimization.

Simulation	U.S. (Gg/month)	Global (Tg/month)
April		
Initial	234	5.67
Optimized	424	6.03
July		
Initial	509	6.37
Optimized	799	6.76
October		
Initial	272	5.69
Optimized	362	5.80

2.2. GEOS-Chem Adjoint Model

[9] The adjoint model is an efficient tool for calculating the gradient of a scalar model response function with respect to all model parameters simultaneously. The adjoint of the GEOS-Chem model was developed specifically for inverse modeling of precursors of inorganic PM_{2.5} with explicit inclusion of gas-phase chemistry, heterogeneous chemistry, and treatment of the thermodynamic couplings of the sulfate-ammonium-nitrate-water aerosol system [Henze *et al.*, 2007, 2009]. As the only adjoint model to explicitly represent this system, it is uniquely capable of assimilating speciated measurements of both gaseous and particulate components using the 4D-Var method [Sandu *et al.*, 2005]. The accuracy of the adjoint model calculations is verified through extensive comparisons of adjoint to finite difference sensitivities. In order to maximize points of comparison between these two approaches, we consider both ensembles of 1-D models (i.e., no horizontal transport) as well as spot tests of the full 3-D adjoint model (testing the full adjoint model for each parameter is prohibitively time consuming, as it would require separate forward model calculations for each of the approximately 10⁵ parameters). Figure 1 shows the results of a week-long test of the sensitivity of surface level NH₃ concentrations in each model column to NH₃ emissions in that column. Sensitivities calculated using the adjoint model are compared with sensitivities calculated using finite differences. All points lie along or near the 1:1 line, demonstrating the accuracy of the adjoint model. Further validation of the adjoint model can be found in previous papers [Henze *et al.*, 2007, 2009, 2012].

2.3. Inverse Modeling

[10] Data assimilation techniques provide a framework for combining observations and models to form an optimal estimation of the state of a system, which in this case is the chemical makeup of the troposphere. To start with, a range of parameters are constructed using control variables, σ , to adjust elements of the vector of model parameters, \mathbf{p} , via application as scaling factors, $p = p_a e^{\sigma}$, where \mathbf{p}_a is the prior parameter estimate. The approach we consider iteratively employs the adjoint of an air quality model

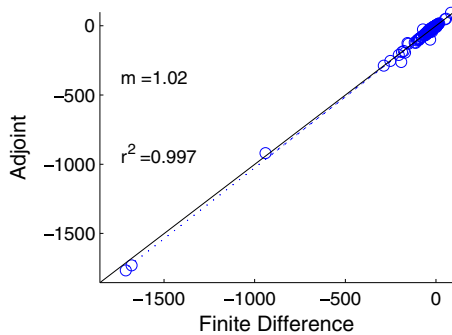


Figure 1. Validation of adjoint model sensitivities via comparison to finite difference (FD) results for week-long simulations. Shown here are the sensitivities [kg] of surface level NH₃ concentrations with respect to NH₃ emissions in an ensemble of column model simulations. Solid lines are 1:1, and dashed lines are regressions with given R^2 and slope m .

in a method referred to as 4D-Var, used here for inverse modeling of emissions. The advantage of this method is that numerous ($O(10^5)$) model parameters can be optimized simultaneously while still retaining the constraints of the full forward model physics and chemistry. This approach to inverse modeling seeks σ that minimizes the cost function, \mathcal{J} , given by

$$\mathcal{J} = \frac{1}{2} \sum_{\mathbf{c} \in \Omega} (H\mathbf{c} - (\mathbf{c}_{obs} - \mathbf{b}))^T \mathbf{S}_{obs}^{-1} (H\mathbf{c} - (\mathbf{c}_{obs} - \mathbf{b})) + \frac{1}{2} \gamma (\sigma - \sigma_a)^T \mathbf{S}_a^{-1} (\sigma - \sigma_a) \quad (1)$$

where H is the observation operator, γ is the regularization parameter, σ_a is the prior estimate of the control variables, \mathbf{S}_a and \mathbf{S}_{obs} are error covariance estimates of the control variables and observations, respectively, Ω is the domain over which observations, \mathbf{c}_{obs} , and model predictions are available, and \mathbf{b} is a bias correction explained in section 5. Overall, the cost function is a specific model response functional, the minimum value of which balances the objectives of improving model performance while ensuring the model itself remains within a reasonable range (as dictated by \mathbf{S}_a^{-1}) of the initial model. Gradients of the cost function with respect to the scaling factors calculated with the adjoint model, $\nabla_{\sigma} \mathcal{J}$, are supplied to an optimization routine (the quasi-Newton L-BFGS-B optimization routine) [Byrd *et al.*, 1995; Zhu *et al.*, 1994], and the minimum of the cost function is sought iteratively. At each iteration, improved estimates of the model parameters are updated and the forward model solution is recalculated.

3. Observations

3.1. Remotely Sensed NH₃ Observations From TES

[11] The high spectral resolution and good signal-to-noise ratio of the TES instrument [Shephard *et al.*, 2008] enabled the first detection of tropospheric ammonia from space with measurements over Southern California and China [Beer *et al.*, 2008]. TES is an infrared Fourier transform spectrometer with spectral resolution of 0.06 cm⁻¹ aboard the NASA Aura satellite, launched 15 July 2004, with a local overpass time of 13:30 and 01:30 [Schoeberl *et al.*, 2006]. TES global survey observations repeat with a 16 day cycle and have a nadir footprint of 5 km × 8 km, for example, that leads to about ~180 daytime retrievals a month over North America after cloud screening (optical depths < 1.0) and applying the TES retrieval quality control flags.

[12] Comparison of model estimates to satellite observations is done via application of the following formula for the TES observational operator, H :

$$H\mathbf{c} = \mathbf{c}_a + \mathbf{A}(\mathbf{M}\mathbf{c} - \mathbf{c}_a) \quad (2)$$

where \mathbf{c} is the model estimated NH₃ profile, \mathbf{M} is a matrix that maps these values to the retrieval units and vertical levels, \mathbf{A} is the averaging kernel, and \mathbf{c}_a is the a priori NH₃ profile used for the retrieval [Shephard *et al.*, 2011]. By comparing TES NH₃ profiles to mapped model estimates, $H\mathbf{c}$, rather than the native model NH₃ profile, \mathbf{c} , the contribution of error in \mathbf{c}_a to the measurement error, \mathbf{S}_{obs} , is minimized [Rodgers, 2000].

[13] For the sake of 2-D visualization, the Representative Volume Mixing Ratio (RVMR) metric [Payne *et al.*, 2009; Shephard *et al.*, 2011] is used to provide a means of comparing TES profiles to model estimates in a manner that accounts for heterogeneity in the instrument's sensitivity to NH₃. RVMR is the average volume mixing ratio within the boundary layer, weighted by a function derived from the retrieval's averaging kernel. This represents a TES sensitivity weighted boundary layer averaged value with the influence of a priori reduced as much as possible [Shephard *et al.*, 2011]. We calculate RVMR only for retrievals that have signal to noise greater than 1 and high thermal contrast. The RVMR is also calculated for the model in those locations which have valid TES retrievals.

3.2. Surface Measurements

[14] In this study, model estimates are evaluated using surface observations of NH₃, sulfate, nitrate, ammonium, and wet deposited NH_x from several monitoring networks throughout the U.S. Surface NH₃ observations are from the National Atmospheric Deposition Program (NADP) Ammonia Monitoring Network (AMoN), which is comprised of triplicate passive ammonia monitoring samplers located at 21 sites across the U.S. with a 2 week long sample accumulation [Puchalski *et al.*, 2011]. All 2 week long observations in each month are averaged to give monthly concentration. The locations of these 21 monitoring stations are shown in Figure 2. Observations from each site are compared with modeled concentrations during the November 2007 through June 2010 period.

[15] Hourly surface NH₃ observations in July of 2008 are from the Southeastern Aerosol Research and Characterization (SEARCH) network [Hansen *et al.*, 2003], which has monitoring stations throughout the Southeast U.S. Different sample frequencies (e.g., daily, 3 day, 6 day, 1 min, 5 min, hourly) are available at different monitoring station. Five minute long observations are available in three stations: Oak Grove, MS; Jefferson Street, GA; and Yorkville, GA. The hourly NH₃ concentration used here is the average of all 5 min long observations of these three stations in each hour.

[16] Wet deposition observations are taken from the NADP National Trends Network (NTN) (<http://nadp.sws.uiuc.edu/NADP>), which are predominantly located away from urban areas and point sources of pollution. NTN has more than 200 sites with week-long sample accumulation.

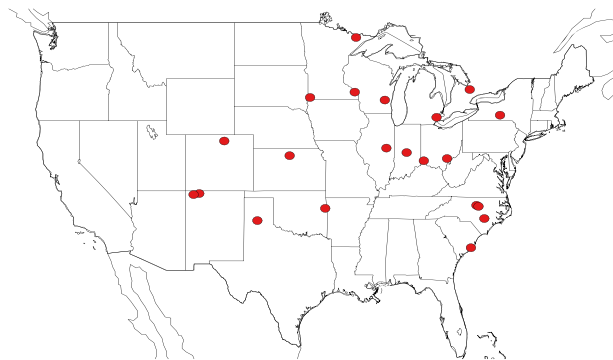


Figure 2. Monitoring site locations for the Ammonia Monitoring Network (AMoN).

[17] Model estimates of sulfate and nitrate aerosol are compared to observations from the Interagency Monitoring of Protected Visual Environments (IMPROVE) network for the year 2008 [Malm *et al.*, 2004]. The IMPROVE network collects PM_{2.5} particles on Teflon, nylon, and quartz filters using a modular, cyclone-based sampler with critical orifice flow control. Sulfate and nitrate aerosols are collected on nylon filters, which are sampled over 24 h every third day.

4. Inversions With Pseudo Data

[18] We first assess the capabilities and limitations of the GEOS-Chem inverse modeling setup in idealized control conditions by designing inverse problems with known solutions. A common framework for testing inverse modeling is the so-called twin experiment in which model simulations are used to generate pseudo observations [Talagrand and P. Courtier, 2008]. In our setup, pseudo observations are generated through application of the TES NH₃ retrieval algorithm [Shephard *et al.*, 2011] to a simulated atmosphere from GEOS-Chem (i.e., radiative transfer calculations are performed to generate pseudo spectra with nominal TES instrument noise, and pseudo NH₃ retrievals produced from these). The standard model emissions used during this simulation are designated as the true emissions. Sampling times, locations, and error estimates reflect those of actual TES observations, although retrieval bias (b) is not included in these tests. For these tests, 87 pseudo TES observations are used from 14 to 19 July 2005, along roughly a dozen global survey transects crossing the midwestern U.S.

[19] To test the inverse model, NH₃ emissions parameters are initialized to values different from the true emissions. In the first test, initial model emissions are half of the true value. Figure 3(a) shows these values in black along with linear line slope m and R^2 . After optimization, the recovered emissions are unbiased and have a visible variance around the true emissions of $\sim 30\%$, as shown in blue in Figure 3(a). In a second test using the same pseudo observations, the model emissions are initially biased high by a factor of 1.8 (Figure 3(b)). The emissions recovered after optimization have a 20% high bias and again a 30% variance about the true values. While the variance of the recovered emissions is similar in both tests, the inversion starting with emissions that are initially too high is less successful.

[20] To further investigate the reasons for this asymmetry and the variance of the optimized emissions, additional tests are performed to separate the possible impacts of inversion error, retrieval bias, and measurement error. In each of the following tests, the true emissions are used to initialize the inversion. The first test uses the same pseudo observations as previously generated. This test again results in a high bias. As the same model state is used to generate the pseudo observations as was used to initialize the inversion, this bias can be attributed to a high bias in the retrieval itself. Retrieval bias, as explained further in Shephard *et al.* [2011], is owing to the fact that the retrieval always selects a moderate or polluted profile as an initial guess in order to avoid the null space of the radiative transfer operator. As the optimal estimation algorithm iterates towards a solution, the process may halt when values reach TES's detection threshold, resulting in a high bias. To test this, the retrieval algorithm is modified to use only a moderate profile as an

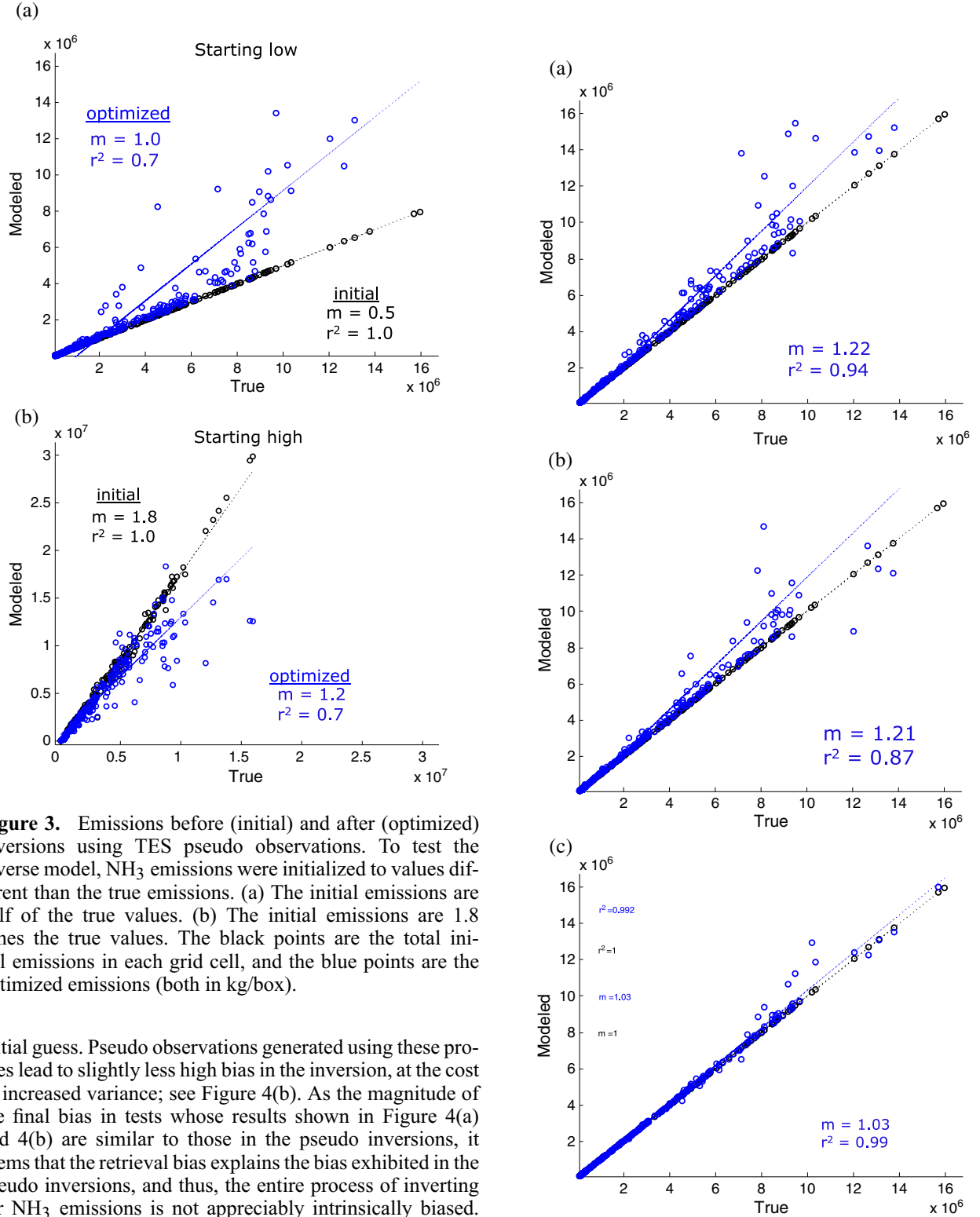


Figure 3. Emissions before (initial) and after (optimized) inversions using TES pseudo observations. To test the inverse model, NH_3 emissions were initialized to values different than the true emissions. (a) The initial emissions are half of the true values. (b) The initial emissions are 1.8 times the true values. The black points are the total initial emissions in each grid cell, and the blue points are the optimized emissions (both in kg/box).

initial guess. Pseudo observations generated using these profiles lead to slightly less high bias in the inversion, at the cost of increased variance; see Figure 4(b). As the magnitude of the final bias in tests whose results shown in Figure 4(a) and 4(b) are similar to those in the pseudo inversions, it seems that the retrieval bias explains the bias exhibited in the pseudo inversions, and thus, the entire process of inverting for NH_3 emissions is not appreciably intrinsically biased. To isolate the impact of measurement noise, the model profiles from the true model are ascribed realistic measurement error and then assimilated. These profiles, unlike the previous tests, correspond directly to the true model and are not retrieved profiles from the retrieval process. The impact of this measurement noise is only a slight adjustment in the emissions. Thus, the variance exhibited in the pseudo inversions is intrinsic to the inversion process itself and would occur even if observations were perfect. This happens

Figure 4. Tests for the possible impacts of inversion error, retrieval bias, and measurement error: (a) retrieval algorithm with a polluted profile as an initial guess; (b) modified retrieval algorithm with a moderate profile as the initial guess; (c) model profiles from the true model were ascribed error of the same size as the measurement error. The black points are the total initial emissions in each grid cell, and the blue points are the optimized emissions (both in kg/box).

because there are variations in emissions that lie in the null space of the forward model. In other words, having some emissions too high and some emissions too low can result in indistinguishable (to TES) distributions of NH₃. Overall, the pseudo observation tests lead us to conclude that (1) measurement noise alone will not lead to unstable inversions, (2) emissions that are underestimated can likely be recovered, (3) emissions that are overestimated will be decreased, though this is countered by bias in the retrievals leading to overestimate of emissions in conditions where the model emissions are initially too high, and (4) that many more iterations and observations would be necessary to reduce the variance of the emissions estimates from the truth, which will be at best $\sim 30\%$. While this variance is substantial, this is a significant improvement over initial errors of $\sim 100\%$. Contrast between points (2) and (3) is likely owing to the larger magnitude of the bias for retrievals with larger values (see Supporting Information Figure S1).

5. TES Assimilation

[21] We next proceed to constrain U.S. NH₃ sources using real observations. TES observations are compared to model estimates from a 2008 GEOS-Chem global $2^\circ \times 2.5^\circ$ simulation using equation (2) to assimilate individual observations. The lifetime of NH₃ is short compared to the residence time of an air mass in one grid cell of our model. As single retrievals may reflect strong sub-grid gradients in NH₃ concentration, we consider satellite observations during four years, 2006–2009, to provide enhanced spatial data coverage for comparison with the 2008 model simulations. Inter-annual comparisons of monthly AMoN NH₃ data indicate no substantial trends in this time period (Figure S3).

[22] Thousands of TES retrievals are available for the assimilation, but not all of the TES retrievals are usable. The satellite cannot always detect NH₃ for several reasons, such as the presence of clouds, low NH₃ concentrations (low signal-to-noise ratio), and poor thermal contrast between the earth and atmosphere. Thus, quality and diagnostic flags are defined to classify and filter the retrievals, keeping only those that have degree of freedom for signal (DOFS) greater than 0.1 or DOFS less than 0.1 but with high (absolute value greater than 7K) thermal contrast. We use the retrievals from daytime only as the retrievals at night are currently being further validated. The TES retrievals are corrected by subtracting mean biases. These biases are generated from the discrepancy between TES retrievals and true profiles [Shephard *et al.*, 2011], and the mean biases are calculated according to the type of the a priori profile (Figure S1). There is separate bias associated with each of the three types of a priori profiles used in the retrieval algorithm. For example, we apply the “unpolluted” bias correction to all observations in which the unpolluted a priori model profile was used in the retrieval algorithm.

[23] A key aspect of inverse modeling is regularization through inclusion of the penalty, or background, term in the cost function, specified through the prior error covariance matrix, S_a , and a regularization parameter, γ . In the absence of rigorous statistical information on the error covariances of the emissions, we assume the errors are uncorrelated and use an L-curve selection criteria [Hansen, 1998] (Figure S2)

to regularize our solution. Still, the relative error for each species is specified as follows. Given the large discrepancies previously noted for NH₃ sources [Henze *et al.*, 2009], uncertainties of NH₃ are taken to be 100% of the maximum NH₃ emissions across the globe. Based on work by other researchers, we assume that the uncertainties in SO₂ [Lee *et al.*, 2011] and NO_x [e.g., Russell *et al.*, 2012] are smaller, conservatively 20% and 50% of the maximum (the impact of these assumed values is assessed in Section 5 and Table 2). With these values, we select the regularization parameter (γ) to be 124 for April, 100 for July, and 50 for October.

[24] TES NH₃ observations are assimilated using the GEOS-Chem adjoint-based inversion. The domain-wide average model NH₃ profiles before and after the inversion are shown along with the average TES profile in Figure 5. The model NH₃ profiles are predominately lower than the TES observations in the prior simulation and have shifted towards the TES profile in the optimized simulation. This leads to reductions of the cost function of 66%, 42%, and 57%, for April, July, and October, respectively. The optimization is considered to have converged when consecutive iterations decrease the cost function by less than 2%. An example of the minimization is provided in Figure S4. While we recognize that further iterations are possible, for practical purposes (each iteration requiring ~ 10 h to compute), this was deemed a sufficient convergence criteria. Limited tests indicate that additional iterations did not drastically alter the results.

[25] The total initial and optimized ammonia emissions are shown in Figure 6. The optimized emissions generally increase over the U.S., with adjustments that are seasonally and spatially heterogeneous. There are large increases in Southern California in all 3 months. Other large increases are located in the central and western U.S., as well as parts of Mexico and Cuba. We do not have much information about NH₃ in Mexico and Cuba due to lack of ground-based measurement records there, but large (e.g., 15 ppb) NH₃ RVMR values are observed in April whereas the corresponding GEOS-Chem model estimates using the initial emissions are very small (e.g., 1.37 ppb). As a result of the inversion, emissions are increased in such areas by up to a factor of 9. Changes to emissions in the East are generally much smaller; emissions in the Atlantic regional are generally unchanged in April and October, while they are higher in the North East in July, and even slightly lower in a few locations in the South.

[26] Figure 7 shows the comparison of NH₃ RVMR from TES and GEOS-Chem before and after the assimilation. There are between 500 and 700 RVMR values in each month including values from 4 years. A linear fit of the model values to the observations is performed in each month, before and after optimization. The slope of this line increases in each month which indicates that most of the RVMR values from GEOS-Chem increase after the optimization. However, the modeled NH₃ RVMRs at low values change only slightly after the optimization in all 3 months. We note however that these differences in RVMR do not reflect observation bias or uncertainty, which contribute to the cost function. In order to show the locations which have significant changes in RVMR, we consider spatial plots of the difference between the TES and GEOS-Chem RVMR before

Table 2. The Effects of A Priori Error Emissions (S_a) on the Optimized Emissions of Different Species. Total Emissions Changes in the U.S. for NH₃, NO_x, and SO₂ When Using Different Values for the Diagonal of $S_a^{\frac{1}{2}}$ in the Optimization. E^0 is the Initial Emissions.

Month	Uncertainties, $S_a^{\frac{1}{2}}/\max(E^0)$			Total emissions changes		
	SO ₂	NO _x	NH ₃	Δ SO ₂	Δ NO _x	Δ NH ₃
April	20%	50%	100%	-11.9%	-14.69%	112.99%
	50%	50%	50%	-11.58%	-15.39%	99.95%
July	20%	50%	100%	-8.35%	-4.24%	54.80%
	50%	50%	50%	-10.23%	-4.58%	53.26%
October	20%	50%	100%	-3.56%	-1.99%	36.07%
	50%	50%	50%	-3.64%	-2.41%	35.16%

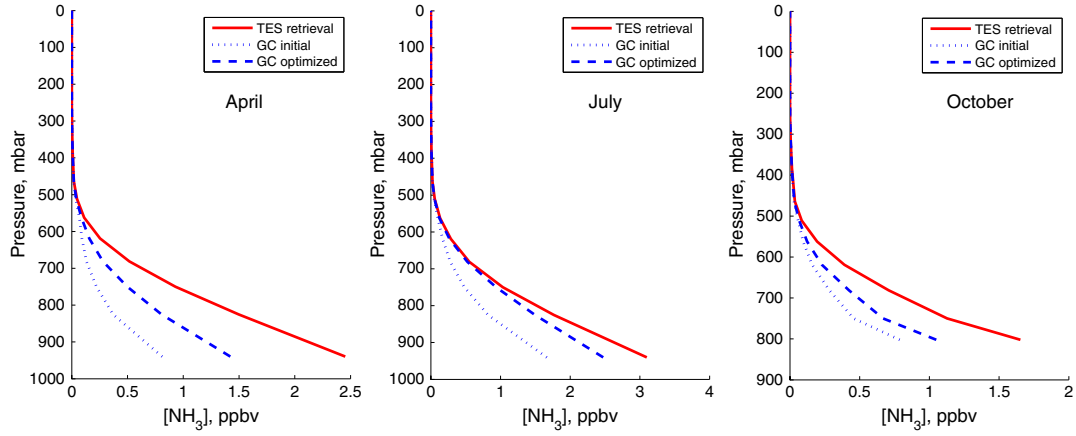


Figure 5. The average of TES retrieval profiles used in the assimilation (solid red line). The average model NH₃ profiles before (blue dotted line) and after (blue dashed line) the assimilation. The TES observation operator is applied to the model profiles.

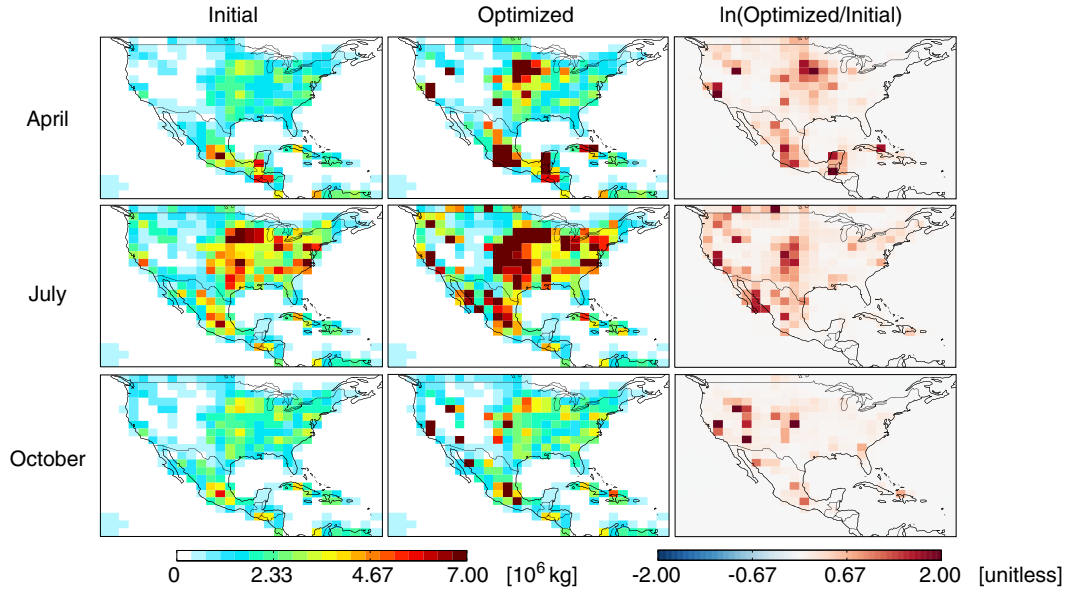


Figure 6. NH₃ emissions from GEOS-Chem before and after the assimilation.

and after the assimilation for each month (Figure 8). Initially, the model RVMRs are generally less than the TES

RVMRs, as indicated by the blue points in the map. After the optimization, model RVMRs increase in many places.

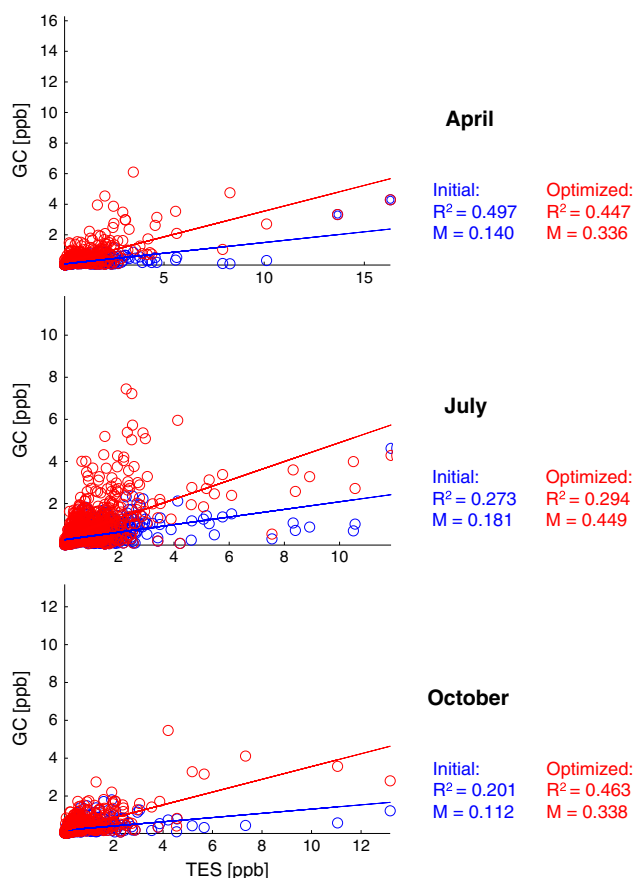


Figure 7. Comparison of NH_3 Representative Volume Mixing Ratio (RVMR) from TES and GEOS-Chem before and after the assimilation.

Some of the model RVMRs are larger than the TES RVMRs, as indicated by the red points in the map, and the overall model bias relative to TES is reduced. The discrepancies between TES RVMRs and model RVMRs change from negative to positive in Southern California and Central U.S. in all 3 months, consistent with the spatial plots showing large increases of ammonia emissions in these locations (Figure 6).

[27] We also assess the sensitivity of these results to the assumed a priori emissions errors, S_a . Table 2 shows the effects of varying a priori errors on the total optimized emissions of different species. We assume the error for NH_3 , $S_a^1(\text{NH}_3)$, to be 50% of the maximum a priori emission for all species. Generally, the results of the inversion are not very different in terms of total emission changes for each species from the base case inversion. However, absolute changes in total emissions of SO_2 and NO_x increase slightly as their uncertainties increase relative to those of NH_3 , while changes in NH_3 total emissions decrease as uncertainty of NH_3 decreases. Differences between the emissions in Table 2 compared to Table 1 stem from the use of an earlier version of the model and TES data set for performing the sensitivity calculations in Table 2; the overall findings are likely still applicable.

6. Posterior Model Evaluation Against Independent Data

[28] In the following sections, we compare output from model simulations using the optimized emissions to independent data sets not used during the inversion. This serves as an important assessment of the robustness of the inverse model solution. Even when the inverse model has converged to a solution in which the model prediction error is minimized and the emissions are consistent with the a priori error assumptions, the validity of the emissions constraints beyond this particular model framework is not guaranteed. The basic 4D-Var approach does not account for bias in the forward chemical transport model, which may significantly impact the top-down emissions constraints. Evaluating against independent data sets is thus critical.

6.1. AMoN and SEARCH

[29] We first consider a comparison of the posterior model results to AMoN NH_3 observations (Figure 9). Initially, the model broadly underestimates AMoN values. After optimization, the NH_3 concentrations increase in each month. The R^2 value increases by 22.4% in April, 29.9% in July, and 27.2% in October. The slope increases by 353.3% in April, 96.1% in July, and 77.1% in October. However, while the root mean square error (RMSE) decreases by 13% in April and 9.5% in October, it increases by 77.6% in July. The normalized mean bias (NMB) after the optimization decreases from -0.678 to -0.069 in April and increases from -0.045 and -0.138 to 0.659 and 0.166 in July and October, respectively. Overall, the model does a better job of capturing the range and variability of NH_3 at AMoN sites in April and October, while in July, the model estimates are biased high.

[30] TES has a detection limit of about 1 ppb and a positive bias of about 0.5 ppb [Shephard et al., 2011]. Model values that are below 1 ppb do not change significantly after the optimization in all 3 months (Figure 9). The bias shown in Figure 9 for July is much higher than 0.5 ppb. One possible reason may be sampling bias due to the TES level of detectability and spatial sampling differences between the TES footprint and the model grid. This is assessed by analyzing NH_3 simulations from high-resolution ($12 \text{ km} \times 12 \text{ km}$) Community Multi-scale Air Quality (CMAQ) model simulations. Surface level NH_3 concentrations throughout the U.S. are compared to concentrations from locations corresponding to successful TES retrievals. The mean surface NH_3 concentration of CMAQ at locations which have successful TES retrievals is about 30% larger than the mean value of that for the whole U.S. This comparison is facilitated by the fact that the TES footprint ($5 \text{ km} \times 8 \text{ km}$) and CMAQ grid cells are similar in size. As shown in Figure 9, changes in large concentrations drive the optimization. A lack of TES observations constraining low values may allow for initial model values that already overestimate low NH_3 concentrations to become even higher in the optimized model, because high TES values, many of which are larger than the initial model estimate, will dominate the cost function. In future work, re-sampling the TES retrievals may be one way to decrease the sampling bias. Increasing the model resolution may also improve our ability to model

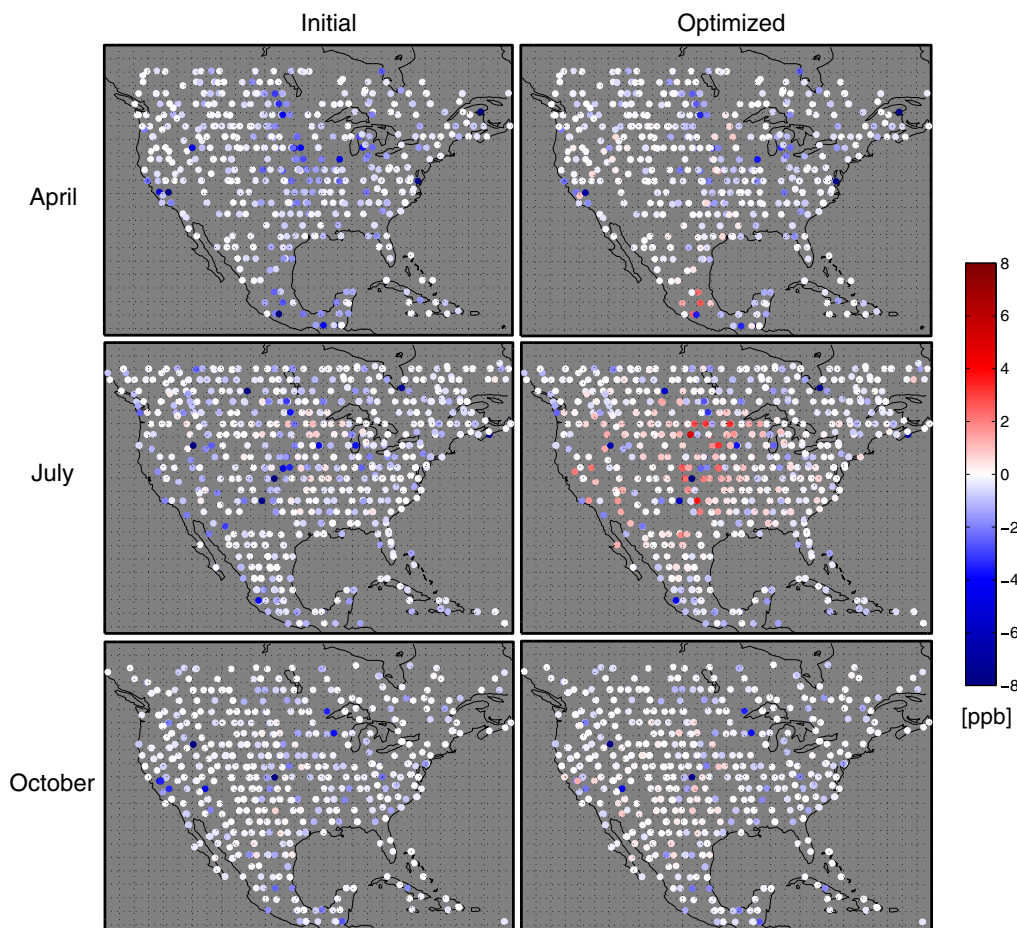


Figure 8. Difference of NH_3 Representative Volume Mixing Ratio (RVMR) between TES and GEOS-Chem before and after the assimilation. The left column shows GEOS-Chem initial RVMR-TES RVMR; the right column shows GEOS-Chem optimized RVMR-TES RVMR.

localized peak NH_3 concentrations measured by TES and to match observations from AMoN.

[31] Lastly, *Jeong et al.* [2013] also compares observations from TES and in situ NH_3 measurements to model simulations, noting similar cases where in situ measurements are lower than the model (CMAQ) simulations, but the TES observations are higher. It is suggested that treatment of the diurnal variability of summertime NH_3 emissions from livestock may play a role in this discrepancy. Comparison of GEOS-Chem hourly NH_3 concentrations to data from the SEARCH network in July paints a similar picture (Figure 10): the constant hourly NH_3 emissions in GEOS-Chem lead to NH_3 concentrations that are overestimated at night and thus also in the monthly average when compared to AMoN. Following *Jeong et al.* [2013], we perform a sensitivity calculation wherein NH_3 emissions from livestock are increased by 90% during the day and reduced by 90% at night. The impact is to decrease the monthly average surface concentrations by several ppb, while increasing daytime boundary layer NH_3 concentrations enough to reduce the cost function by 25%. In addition, *Jeong et al.* [2013] also show that bi-directional exchange of NH_3 in July leads to increased and decreased emissions in different regions (10% overall increase), yet NH_3 concentrations increase everywhere. This is in contrast to April

and October, where emissions and concentrations decrease. Physical mechanisms have thus been identified by which mid-day model profiles of NH_3 may be increased without large increases in emissions. Further consideration of mechanistic, process-based treatment of NH_3 sources in global models is therefore warranted.

6.2. NTN

[32] As an additional check of the broad NH_x budget, we consider the NH_x wet deposition as recorded by NTN (NADP) sites. To make this comparison, we consider that simulated precipitation is a critical driver in the performance of the GEOS-Chem-simulated wet deposition estimates, as biases in the model estimated precipitation can lead to biases in the GEOS-Chem model estimates. We therefore adjust the modeled wet deposition diagnostic to account for differences in the modeled and observed precipitation by linearly scaling the model estimated wet deposition by the ratio of the observed to estimated precipitation.

[33] Figure 11 shows the comparison of modeled wet deposition with the NTN observations. Generally, the inversion increases wet deposition during all 3 months. Also, the square of the correlation coefficient (R^2) improves in each month: 11.4% in April, 14.1% in July, and 10.3% in October. In April and October, optimized values compare

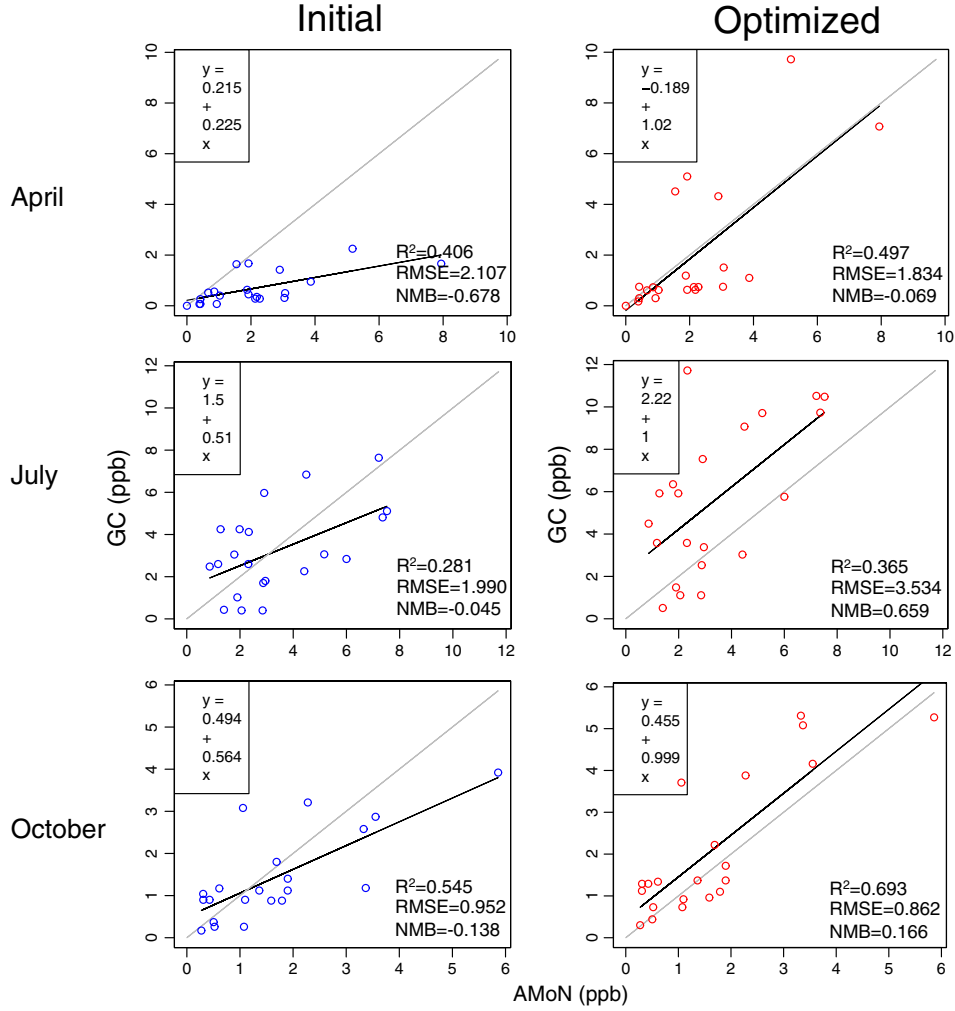


Figure 9. Comparison of GEOS-Chem NH_3 concentrations with observations from AMoN sites before and after the assimilation. The square of the correlation coefficient (R^2), root mean square error (RMSE), and normalized mean bias (NMB) are shown. Black solid lines are regressions. Gray dashed lines are 1:1.

better with the NTN observations. In contrast, the slope of the linear regression line increases from 1.08 to 1.59 in July. This is consistent with the high bias of the inversion relative to the AMoN surface observations of NH_3 in July. Comparisons between GEOS-Chem and NTN observations are also shown in *Zhang et al.* [2012]. They compare the NH_3 wet deposition from GEOS-Chem at the $0.5^\circ \times 0.67^\circ$ resolution with NTN observations from 2006. Differences between this study and the present work are the model resolution, data filtering, the number of months per season included in the comparisons (one versus three), and the emission inventories in *Zhang et al.*, [2012], which include domain-wide seasonal adjustments based on in situ observations to minimize seasonal bias.

6.3. IMPROVE

[34] For an additional evaluation, we also compare the assimilated results with aerosol observations from IMPROVE in 2008. Figure 12 shows that model optimization slightly decreases the sulfate concentrations and increases the nitrate concentrations, which facilitates increases in NH_3 concentrations to match TES observations.

Still, the changes are small, and the sulfate concentration from the model has a reasonable correlation with the IMPROVE observation before and after optimization in each month. Note that the outlier in October that has a large observed value but a nearly zero model value is located in Hawaii. The model does not represent this high value owing to the proximity of the observation to the local volcanic source.

[35] Optimized NH_3 emissions do not help the comparison of simulated nitrate to IMPROVE observations, which initially are over predicted. The balance of sulfate and nitrate from IMPROVE sites alone implies that NH_3 sources are too high [*Henze et al.*, 2009]. However, model NH_3 in the present work increases in order to improve agreement with TES NH_3 , leading to more nitrate formation, and the nitrate bias becomes even higher compared to IMPROVE. It is thus a challenge to resolve the underestimates of boundary layer NH_3 with the overestimates of nitrate. HNO_3 formation in the model is perhaps excessive [*Zhang et al.*, 2012], and overly shallow nighttime boundary layers may contribute to enhanced nitrate [*Heald et al.*, 2012]. We perform additional sensitivity studies, reducing the heterogeneous uptake coef-

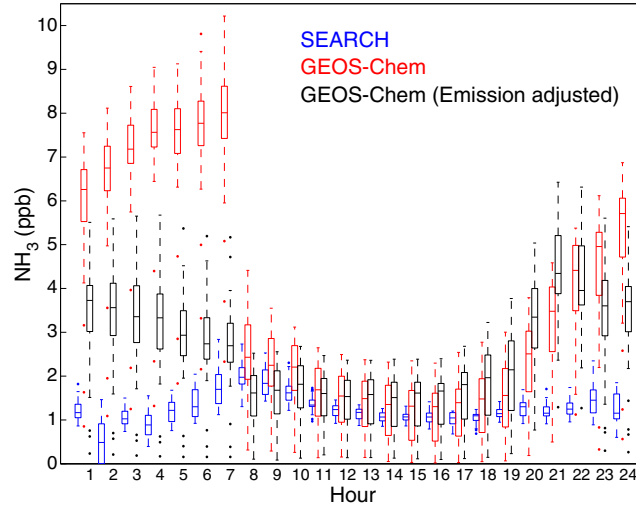


Figure 10. Comparison of initial (un-optimized) GEOS-Chem hourly NH_3 concentrations with observations from SEARCH (blue) sites in July. The red color represents the hourly NH_3 concentrations with the standard constant hourly NH_3 emissions in GEOS-Chem. The black color represents the hourly NH_3 concentrations with NH_3 livestock emissions increased by 90% during the day and reduced by 90% at night.

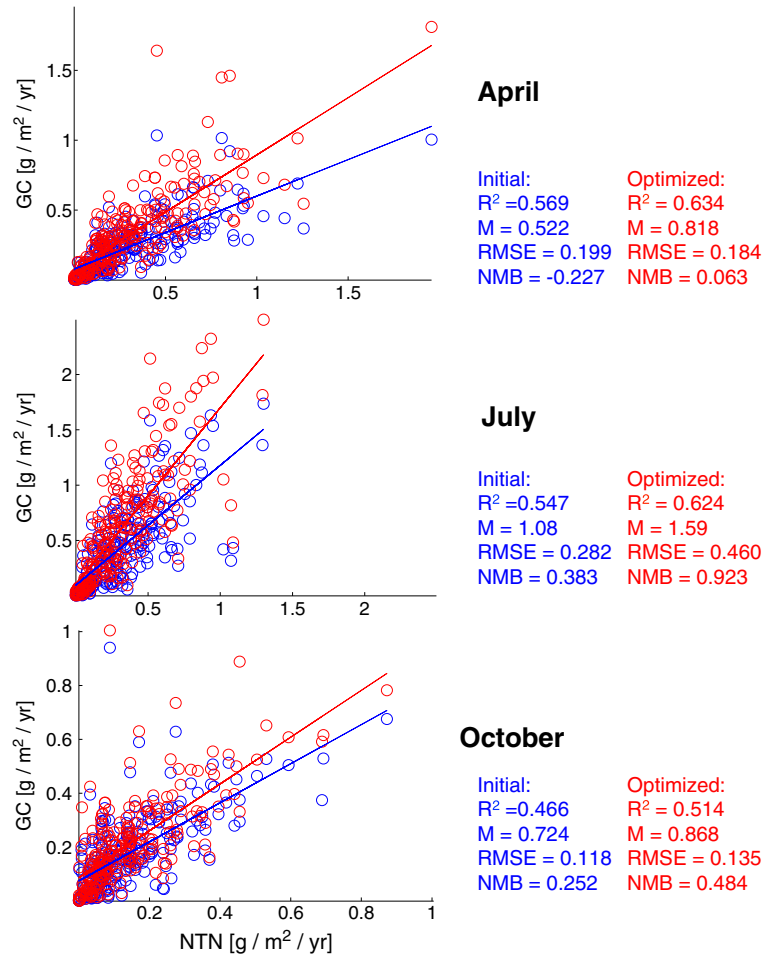


Figure 11. Comparison of GEOS-Chem NH_x wet deposition with observations from NTN (NADP) sites before (blue) and after (red) the assimilation.

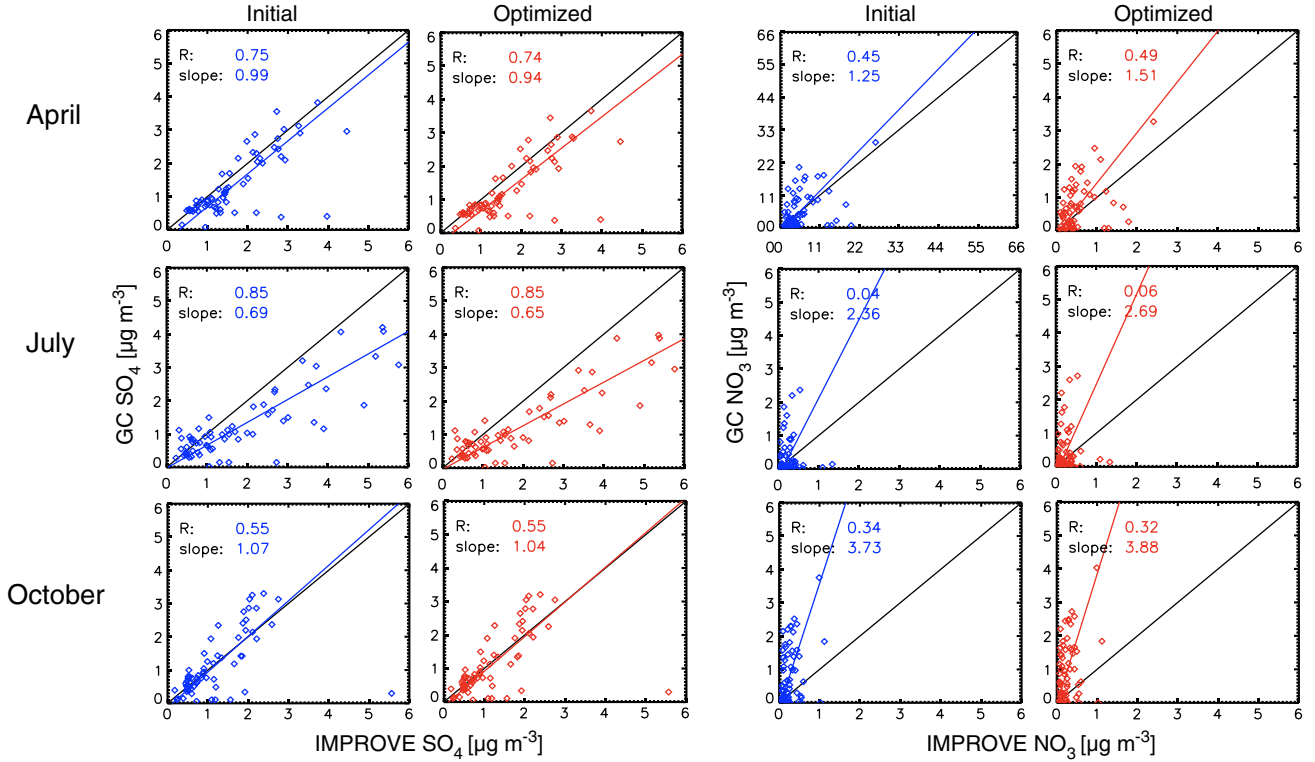


Figure 12. Comparison of GEOS-Chem SO₄ and NO₃ concentrations with observations from IMPROVE sites before and after the assimilation.

ficient for N₂O₅ hydrolysis by an order of magnitude and adjusting the boundary layer height following *Heald et al.* [2012]. While such changes do impact the nitrate simulation, the nitrate high bias persists, and the comparison of the base model NH₃ to the AMoN and TES observations was not notably affected. Enforcing a diurnal variation in NH₃ livestock emissions decreased average surface nitrate concentrations by up to a $\mu\text{g}/\text{m}^3$, as the GEOS-Chem overestimates of nighttime NH₃ likely contributed to excessive partitioning of HNO₃ at night. Thus, to achieve closure relative to all data sets, it is evident that assessment of model error beyond NH₃ sources, in terms of scavenging efficiencies, deposition, diurnal variability, and further investigation of HNO₃ production is required.

7. Conclusions

[36] Here we have considered the potential for space-based observations of NH₃ to constrain monthly average emissions. Initial tests using pseudo observations show that under ideal conditions (i.e., a perfect model) using 2 weeks worth of TES data, 70% of the variance of the emissions can be constrained in terms of total magnitude. We then proceed to assimilate TES observations in April, July, and October for multiple years. We present a range of constrained prediction results and evaluate them with independent data sets. Generally, model optimization increases NH₃ concentrations and NH_x wet deposition. Overall, the model does a better job of capturing the range and variability of NH₃ at AMoN sites in April and October, while in July, the model estimates are consistently biased high. Compared to the wet deposition observations of NTN, optimization decreases the

normalized mean bias (NMB) in April, enhances the NMB in July and October, but overall leads to increased correlation of modeled and observed values. Modeled SO₄²⁻ aerosol concentrations slightly decrease, and NO₃⁻ aerosols concentrations increase, which increases the bias compared to the IMPROVE observations.

[37] Overall, remote sensing constraints indicate that the initial NH₃ emissions inventory appears to be broadly underestimated in several areas throughout the U.S., particularly in the West. This is consistent with recent works regarding NH₃ levels in California throughout the year [*Walker et al.*, 2012] and the U.S. in the spring [*Heald et al.*, 2012], as well as in situ measurements in April and October. Still, the absolute extent of the emissions underestimation in these seasons is still in question, as the precise accuracy of the satellite observations is difficult to specify, model resolution is not matched to the satellite resolution, and model processes could have error contributing to uncertainty in the inversion. A greater fraction of peak values are included in the assimilation owing to satellite detection limits, leading to a sampling bias that may cause the inverse model to overestimate emissions in July, as indicated by high biases in the optimized model simulations of surface level NH₃ concentrations and NH_x deposition. Future work will apply a newly developed higher resolution (e.g., $0.5^\circ \times 0.67^\circ$) version of the inverse model to further investigate this issue.

[38] Additional sensitivity studies, comparisons to hourly observations from SEARCH, and the recent regional modeling work of *Jeong et al.* [2013] indicate that diurnal variability of livestock emissions and bi-directional flux from fertilizer sources may play important roles in resolving discrepancies between average surface level concentrations

and mid-day profiles of NH₃ and may also impact the high bias of nitrate in GEOS-Chem [Heald et al., 2012]. Further assessment of mechanistic, process-based treatment of NH₃ sources in global models is thus needed. Additional measurements, from either expanded in situ monitoring networks or geostationary remote sensing instruments, would greatly serve this need.

[39] **Acknowledgments.** This work is supported by NASA grants NNX09AN77G and NNX10AG63G and EPA STAR award RD834559. While this manuscript has been reviewed by the Environmental Protection Agency and approved for publication, it may not reflect official agency views or policies.

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